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(11) EP 1 000 937 A1

(12)

EUROPEAN PATENT APPLICATION

published in accordance with Art. 158(3) EPC

(43) Date of publication:

17.05.2000 Bulletin 2000/20

(21) Application number: 98919595.3

(22) Date of filing: 14.05.1998

(51) Int. Cl.⁷: C07D 263/24, C07D 263/44

(86) International application number:
PCT/JP98/02129

(87) International publication number:
WO 99/02508 (21.01.1999 Gazette 1999/03)

(84) Designated Contracting States:
CH DE FR GB IT LI

(30) Priority: 10.07.1997 JP 18515097

(71) Applicant:
UBE INDUSTRIES LIMITED
Ube-shi, Yamaguchi 755-8633 (JP)

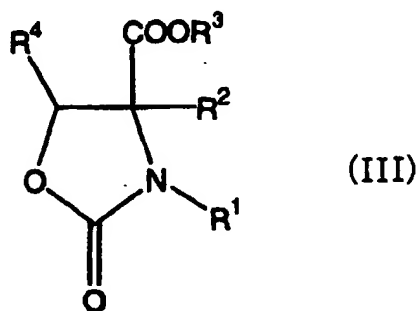
(72) Inventors:
• MIYATA, Hiroyuki
Ube Research Laboratory
Ube-shi Yamaguchi 755-8633 (JP)

• SATAKE, Nobuya
Ube Research Laboratory
Ube-shi Yamaguchi 755-8633 (JP)
• HONMA, Takashi
Ube Research Laboratory
Ube-shi Yamaguchi 755-8633 (JP)
• ATAKA, Kikuo
Ube Research Laboratory
Ube-shi Yamaguchi 755-8633 (JP)

(74) Representative: HOFFMANN - EITL
Patent- und Rechtsanwälte
Arabellastrasse 4
81925 München (DE)

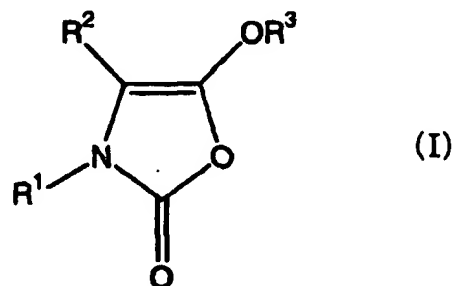
(54) PROCESS FOR PRODUCING 4-ALKOXYCARBONYL-2-OXAZOLIDINONE COMPOUNDS

(57) The present invention relates to a process for producing 4-alkoxycarbonyl-2-oxazolidinone compound represented by the formula (III):



acetyl group or a benzoyl group,

which comprises reacting a 5-alkoxy-2(3H)oxazolone compound represented by the formula (I):



wherein R¹ represents H, an alkyl group, an cycloalkyl group, an alkenyl group or a phenyl group, R² represents H, an alkyl group, a phenyl group or an alkenyl group, R³ represents an alkyl group, a cycloalkyl group, an alkenyl group, or a phenyl group, and R⁴ represents H, an alkyl group, an alkenyl group, a cycloalkyl group, an alkynyl group, an aryl group, a 5- or 6-membered heteroaromatic ring group, an alkoxycarbonyl group, an

wherein R¹, R² and R³ have the same meanings as defined above,

and an aldehyde compound represented by the formula (II):



wherein R⁴ has the same meaning as defined

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above, in the presence of a Lewis acid catalyst.

Description**Technical field**

5 [0001] This invention relates to a process for producing a 4-alkoxycarbonyl-2-oxazolidinone compound which is useful as synthetic starting materials or intermediates of pharmaceuticals and agricultural chemicals.

[0002] 4-Alkoxycarbonyl-2-oxazolidinone compound can be easily led to a β -hydroxy- α -amino acid compound which is useful as pharmaceuticals, or pharmaceutical intermediates or starting materials in accordance with the method described in, for example, Journal of American Chemical Society, 118, pp. 3584-3590 (1996) by subjecting to
10 alkali hydrolysis using an aqueous potassium hydroxide solution.

Background art

[0003] As the conventional method for preparing 4-alkoxycarbonyl-2-oxazolidinone compound, the following meth-
15 ods have been known.

① : In Nihon Kagakukai Zasshi, vol. 82, p. 1075, (1961), there is disclosed a method in which phosgene is acted on D,L-threonine in the presence of sodium hydroxide, and esterifying with methanol and hydrochloric acid to obtain 5-methyl-4-methoxycarbonyl-2-oxazoline. This method is not industrially sufficient method in the point of
20 using β -hydroxy- α -amino acid compound which is difficultly obtained as a starting material.

② : In Nihon Kagakukai Zasshi, vol. 82, p. 1075, (1961), there is disclosed a method in which N-Cbz-D,L-arothre-
onine is subjected to ring opening in the presence of sodium hydroxide, and esterifying with methanol and hydro-
chloric acid to obtain 5-methyl-4-methoxycarbonyl-2-oxazolidinone. This method is also not industrially sufficient
25 method in the point of using, as a starting material, β -hydroxy- α -amino acid compound the starting material of
which is difficultly obtained.

③ : In Journal of Organic Chemistry, vol. 44, p. 3967, (1979), there is disclosed a method in which N-carbobenzy-
loxyglycine ethyl ester is reacted with benzaldehyde in the presence of lithium diisopropylamide to obtain 5-phenyl-
4-methoxycarbonyl-2-oxazolidinone. However, this method is not industrially satisfied preparation method in the
point that lithium diisopropylamide handling of which is difficult must be used.

④ : In Tetrahedron Letters, vol. 29, p. 2069 (1988), there is disclosed a method of obtaining 3-methyl-5-(1-methyl-
3-pentenyl)-4-methoxycarbonyl-2-oxazolidinone by reacting 3-(1-methyl-3-pentenyl)-2,3-epoxy-1-propanol and
methyl isocyanate to obtain 4-hydroxymethyl-oxazolidine and oxidizing the resulting compound with a chromium
compound and esterifying with diazomethane. However, the method is not industrially sufficient method in the
points that the steps are long and a chromium compound or diazomethane which is difficultly used industrially must
35 be used.

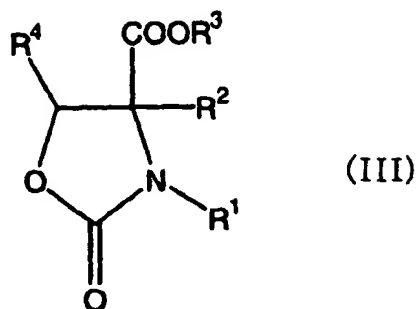
[0004] Accordingly, in the above-mentioned conventionally known methods ①, ②, ③ and ④, there are problems
that the starting materials are difficultly obtained, a starting material which is difficultly used for industrial purpose must
be used, the steps are long and yield is low, so that a novel preparation method of 4-alkoxycarbonyl-2-oxazolidinone
40 compound which can overcome these problems has been desired.

[0005] An object of the present invention is to provide a preparation method of a 4-alkoxycarbonyl-2-oxazolidinon
compound which is available as a synthetic starting material or an intermediate of medicine or agricultural chemicals.

Disclosure of the invention

45 [0006] The present inventors have investigated to solve the above-mentioned problems and as a result, they have
found that by reacting 5-alkoxy-2(3H)oxazolone compound which has not yet been described in the references and an
aldehyde compound in the presence of a Lewis acid catalyst, a 4-alkoxycarbonyl-2-oxazolidinone compound can be
obtained easily with high yield to accomplish the present invention. That is, the present invention is as mentioned below.

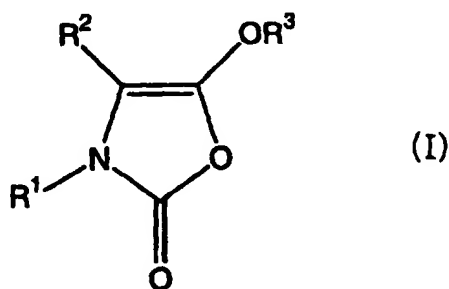
50 [0007] That is, the present invention related to a process for producing a 4-alkoxycarbonyl-2-oxazolidinone com-
pound represented by the formula (III):



15 wherein R¹ represents a hydrogen atom, a C₁ to C₁₀ alkyl group which may be substituted, a C₃ to C₁₀ cycloalkyl group which may be substituted, a C₂ to C₁₀ alkenyl group which may be substituted or a phenyl group which may be substituted, R² represents a hydrogen atom, a C₁ to C₁₀ alkyl group which may be substituted, a phenyl group which may be substituted or a C₂ to C₁₀ alkenyl group which is not substituted, R³ represents a C₁ to C₁₀ alkyl group which may be substituted, a C₃ to C₁₀ cycloalkyl group which may be substituted, a C₂ to C₁₀ alkenyl group which may be substituted (provided that a 2-alkenyl group is excluded), or a phenyl group which may be substituted, and R⁴ represents a hydrogen atom, a C₁ to C₂₀ alkyl group which may be substituted, a C₂ to C₂₀ alkenyl group which may be substituted, a C₃ to C₁₀ cycloalkyl group which may be substituted, a C₂ to C₂₀ alkynyl group which may be substituted, a C₆ to C₂₀ aryl group which may be substituted, a 5- or 6-membered heteroaromatic ring group having 1 or 2 hetero atoms selected from N, O and S which may be substituted, a C₁ to C₆ alkoxycarbonyl group which may be substituted, an acetyl group or a benzoyl group, which comprises reacting a 5-alkoxy-2(3H)oxazolone compound represented by the formula (I):

20

25



40 wherein R¹, R² and R³ have the same meanings as defined above,

45 and an aldehyde compound represented by the formula (II):



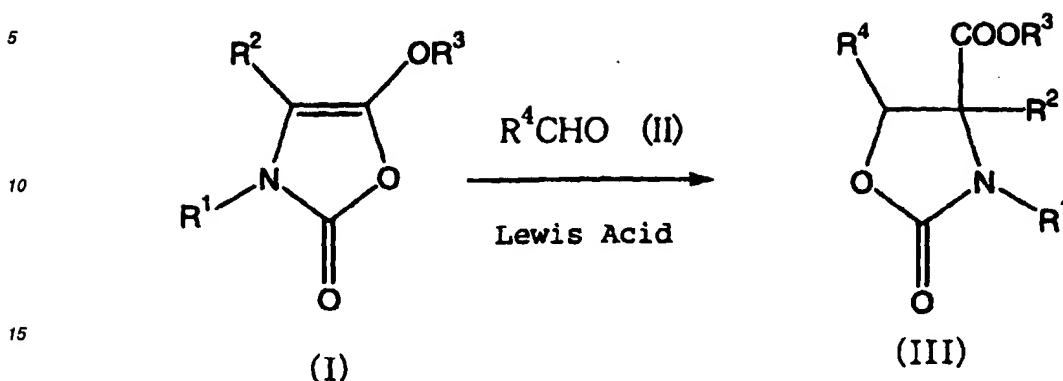
50 wherein R⁴ has the same meaning as defined above, in the presence of a Lewis acid catalyst to obtain the title compound.

Best mode for carrying out the invention

[0008] In the following, the present invention is explained in detail.

55 [0009] The process of the present invention can be shown, for example by the following reaction formula (I):

Reaction formula (I)



20 wherein R^1 , R^2 , R^3 and R^4 have the same meanings as defined above.

[0010] In the present invention, R^1 in the compound (I) represented by the formula (I) represents a hydrogen atom, a C_1 to C_{10} alkyl group which may be substituted, a C_3 to C_{10} cycloalkyl group which may be substituted or a C_2 to C_{10} alkenyl group which may be substituted or a phenyl group which may be substituted.

[0011] "The C_1 to C_{10} alkyl group which may be substituted" represented by R^1 means (1) "a C_1 to C_{10} alkyl group having no substituent" or (2) "a C_1 to C_{10} alkyl group having a substituent(s)".

[0012] As "the C_1 to C_{10} alkyl group having no substituent" of (1), there may be mentioned, for example, a straight or branched C_1 to C_{10} alkyl group such as a methyl group, an ethyl group, a propyl group (including an isomer), a butyl group (including isomers thereof), a pentyl group (including isomers thereof), a hexyl group (including isomers thereof), a heptyl group (including isomers thereof), an octyl group (including isomers thereof), a nonyl group (including isomers thereof) or a decyl group (including isomers thereof), etc.

[0013] As the substituent for "the C_1 to C_{10} alkyl group having a substituent(s)", there may be mentioned, for example, a cyano group, a benzyloxy group, a phthalimino group, an acylamino group, an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion, a trialkylsilyloxy group having the same or different straight or branched C_1 to C_6 alkyl group portion, a halogen atom, an aryl group which may be substituted, "a 5- or 6-membered heteroaromatic ring group (hereinafter also referred to as "a heteroaromatic ring group") containing 1 or 2 hetero atoms selected from N, O and S", or a C_1 to C_{10} alkoxy group which may be substituted. Incidentally, the number of the substituent(s) and the position thereof is not limited.

[0014] "The aryl group which may be substituted" as the substituent for "the C_1 to C_{10} alkyl group having a substituent(s)" of (2) means (2-1) "an aryl group having no substituent" or (2-2) "an aryl group having a substituent(s)".

[0015] As "the aryl group having no substituent" of (2-1), there may be mentioned, for example, a phenyl group, a naphthyl group, an anthryl group or a phenanthryl group, etc.

[0016] As "the aryl group having a substituent(s)" of (2-2), there may be mentioned, for example, a nitro group; a cyano group; a benzyloxy group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different, straight or branched C_1 to C_6 alkyl group portion; a halogen atom; a straight or branched C_1 to C_6 alkyl group; or a straight or branched C_1 to C_6 alkoxy group. Incidentally, the number and the position of the substituent are not limited.

[0017] "The heteroaromatic ring group which may be substituted" as a substituent of "the C_1 to C_{10} alkyl group having a substituent(s)" of (2) means (2-3) "a heteroaromatic ring group having no substituent" or (2-4) "a heteroaromatic ring group having a substituent(s)".

[0018] As (2-3) "the heteroaromatic ring group having no substituent", there may be mentioned, for example, a furyl group, a thienyl group, a pyrrolyl group, a 2H-pyrrolyl group, an oxazolyl group, an isoxazolyl group, a thiazolyl group, an isothiazolyl group, an imidazolyl group, a pyrazolyl group, a pyranil group, a pyridyl group, a pyridazyl group, a pyrimidyl group or a pyrazinyl group, etc.

[0019] As (2-4) "the heteroaromatic ring group having a substituent(s)", there may be mentioned, for example, a nitro group; a cyano group; a benzyloxy group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different, straight or branched C_1 to C_6 alkyl group portion; a

halogen atom; a straight or branched C₁ to C₆ alkyl group; or a straight or branched C₁ to C₆ alkoxy group. Incidentally, the number and the position of the substituent are not limited.

[0020] As (2-5) "the C₁ to C₁₀ alkoxy group having no substituent", there may be mentioned, for example, a straight or branched C₁ to C₁₀ alkoxy group such as a methoxy group, an ethoxy group, a propoxy group (including an isomer), a butoxy group (including isomers thereof), a pentyloxy group (including isomers thereof), a hexyloxy group (including isomers thereof), a heptyloxy group (including isomers thereof), an octyloxy group (including isomers thereof), a nonyloxy group (including isomers thereof) or a decyloxy group (including isomers thereof), etc.

[0021] As the substituent for (2-6) "the C₁ to C₁₀ alkoxy group having a substituent(s)", there may be mentioned, for example, a benzyloxy group; a phenoxy group; a methoxyethoxy group; a trialkylsilyloxy group having the same or different, straight or branched C₁ to C₆ alkyl group portion or a straight or branched C₁ to C₆ alkoxy group. Incidentally, the number and the position of the substituent are not limited.

[0022] "The cycloalkyl group which may be substituted" represented by R¹ in the compound (I) represented by the formula (I) means (3) "a C₃ to C₁₀ cycloalkyl group having no substituent" or (4) "a C₃ to C₁₀ cycloalkyl group having a substituent(s)".

[0023] As (3) "the C₃ to C₁₀ cycloalkyl group having no substituent", there may be mentioned, for example, a C₃ to C₁₀ cycloalkyl group such as a cyclopropyl group, a cyclobutyl group, a cyclopentyl group, a cyclohexyl group, a cycloheptyl group, a cyclooctyl group, a cyclononyl group, a cyclodecyl group, a bornyl group or an adamantyl group, etc.

[0024] As the substituent for (4) "the C₃ to C₁₀ cycloalkyl group having a substituent(s)", there may be mentioned, for example, a cyano group; a benzyloxy group; a benzosulfonyl group; an acylamino group; an alkoxycarbonyl group having a straight or branched C₁ to C₆ alkyl group portion; a trialkylsilyloxy group having the same or different, straight or branched C₁ to C₆ alkyl group portion; a halogen atom; a straight or branched C₁ to C₆ alkyl group; an aryl group which may be substituted or a straight or branched C₁ to C₆ alkoxy group. Incidentally, the number and the position of the substituent are not limited.

[0025] "The C₂ to C₁₀ alkenyl group which may be substituted" represented by R¹ in the compound (I) represented by the formula (I) has the same meanings as (5) "a C₂ to C₁₀ alkenyl group having no substituent" or (6) "a C₂ to C₁₀ alkenyl group having a substituent(s)".

[0026] As (5) "the C₂ to C₁₀ alkenyl group having no substituent", there may be mentioned, for example, a straight or branched C₂ to C₁₀ alkenyl group such as an ethenyl group, a propenyl group (including its isomer), a butenyl group (including isomers thereof), a pentenyl group (including isomers thereof), a hexenyl group (including isomers thereof), a heptenyl group (including isomers thereof), an octenyl group (including isomers thereof), a nonenyl group (including isomers thereof) or a decenyl group (including isomers thereof), etc.

[0027] As the substituent for (6) "the C₂ to C₁₀ alkenyl group having a substituent(s)", there may be mentioned, for example, a cyano group; a benzyloxy group; an alkoxycarbonyl group having a straight or branched C₁ to C₆ alkyl group portion; a trialkylsilyloxy group having the same or different, straight or branched C₁ to C₆ alkyl group portion; a halogen atom; an aryl group which may be substituted; a heteroaromatic ring group which may be substituted or a straight or branched C₁ to C₆ alkoxy group. Incidentally, the number of the substituent and the position thereof are not particularly limited.

[0028] "The phenyl group which may be substituted" represented by R¹ in the compound (I) represented by the formula (I) means a phenyl group or "a phenyl group having a substituent(s)". As the substituent for "the phenyl group having a substituent(s)", there may be mentioned, for example, a straight or branched C₁ to C₆ alkyl group, a nitro group, a benzyloxy group, a halogen atom, an acylamino group and a straight or branched C₁ to C₆ alkoxy group. Incidentally, the number and the position of the substituent(s) are not limited.

[0029] Specific examples of such R¹ may include, for example, a hydrogen atom, a methyl group, an ethyl group, a propyl group, an isopropyl group, a butyl group, an isobutyl group, a secbutyl group, a tert-butyl group, a pentyl group, a hexyl group, a heptyl group, an octyl group, a nonyl group, a decyl group, a 2-cyanoethyl group, a 1-methoxycarbonylethyl group, a 2-trimethylsilyloxyethyl group, a 2-benzyloxyethyl group, a 2-trifluoroethyl group, a 2-chloroethyl group, a benzyl group, a p-nitrobenzyl group, a cyanobenzyl group, a p-methoxycarbonylbenzyl group, a p-trimethylsilyloxybenzyl group, a p-benzyloxybenzyl group, a 3,4-difluorobenzyl group, a p-methylbenzyl group, an o-methylbenzyl group, a 3,4-dichlorobenzyl group, an o-fluorobenzyl group, a p-fluorobenzyl group, a 2,4-dimethylbenzyl group, a p-isopropylbenzyl group, a p-tert-butylbenzyl group, an o-methoxybenzyl group, a m-methoxybenzyl group, a p-methoxybenzyl group, a 3,4-dimethoxybenzyl group, a 2,4-dimethoxybenzyl group, a 2-ethoxybenzyl group, a p-isopropoxybenzyl group, a p-tert-butoxybenzyl group, a 1-phenylethyl group, a 1-(p-nitrophenyl)ethyl group, a 1-(4-bromophenyl)ethyl group, a 1-(4-fluorophenyl)ethyl group, 1-(p-methoxyphenyl)ethyl group, a 1-(4-chlorophenyl)ethyl group, a 1-(1-naphthyl)ethyl group, a 1-(2-naphthyl)ethyl group, a diphenylmethyl group, a di(4-chlorophenyl)methyl group, a di(4-methoxyphenyl)methyl group, a trityl group, a 1-(2-phenanthryl)ethyl group, a 1-(9-anthranlyl)ethyl group, a furfuryl group, a 2-thienylmethyl group, a 2-pyridylmethyl group, a 3-pyridylmethyl group, an isothiazolylmethyl group, a 2-pyrazolylethyl group, a (2H-pyrrolyl) group, an (N-methylpyrrolyl)methyl group, an isoxazolylmethyl group, a 2-methoxyethyl group, a

2-ethoxyethyl group, a 2-(n-propoxy)ethyl group, a 2-isopropoxyethyl group, a 3-(n-butoxy)propyl group, 2-(sec-butoxy)ethyl group, a 2-(tert-butoxy)ethyl group, a 2-hexyloxyethyl group, a 2-methoxy-n-butyl group, a 2-(tert-butoxy)-1,1-dimethyl-ethyl group, a 2-octyloxyethyl group, a 2-nonyloxyethyl group, a 2-heptyloxyethyl group, a 2-methoxyethoxyethyl group, a 2-(benzyloxymethoxy)-ethyl group, a 2-(2-methoxyethoxy-methoxy)ethyl group, a 2-(ethoxymethoxy)-ethyl group, a 2-(phenoxymethoxy)-ethyl group, a 2-formaminoethyl group, a 2-acetaminoethyl group, a 2-chloroacetaminoethyl group, a 2-benzoylaminoethyl group, a 2-phenylacetaminoethyl group, a 2-methoxycarbonylaminoethyl group, a 2-ethoxycarbonylaminoethyl group, a 2-allyloxycarbonylaminoethyl group, a 2-tert-butoxycarbonylaminoethyl group, a 2-benzyloxycarbonylaminoethyl group, an ethynyl group, a propenyl group, a butenyl group, a pentenyl group, a hexenyl group, a heptenyl group, an octenyl group, a nonenyl group, a decenyl group, a 2-cyanoethynyl group, a 1-methoxycarbonyl-2-propenyl group, a 1-trimethylsilyloxymethyl-2-propenyl group, a 1-benzyloxymethyl-2-propenyl group, a cinnamyl group, a 2-methoxymethyl-2-propenyl group, a 2-ethoxymethyl-2-propenyl group, a cyclopropyl group, a cyclobutyl group, a cyclopentyl group, a cyclohexyl group, a cycloheptyl group, a cyclooctyl group, a cyclononyl group, a cyclodecyl group, a 1-methoxycarbonylcyclopropyl group, a 2-phenylcyclopropyl group, a 2-fluorocyclopropyl group, a 2-chlorocyclopropyl group, a 2-benzyloxycyclopropyl group, a 1-cyanocyclopentyl group, a 2-norbornyl group, a bornyl group, a 1-adamantyl group, a 4-methylcyclohexyl group, a 2-methylcyclohexyl group, a 2,3-dimethylcyclohexyl group, a 1-methoxycarbonylcyclohexyl group, a 2-methoxycyclohexyl group, a 2-trimethylsilyloxycyclohexyl group, a 2-benzyloxycyclohexyl group, a 4-benzyloxycyclohexyl group, 4-tert-butylcyclohexyl group, a menthyl group, a 8-phenylmenthyl group, a phenyl group, a 2-methylphenyl group, a 3-methylphenyl group, a 4-methylphenyl group, a 4-tert-butylphenyl group, a 3,4-dimethylphenyl group, a 4-ethylphenyl group, a 2-chlorophenyl group, a 3-chlorophenyl group, a 4-chlorophenyl group, a 2-fluorophenyl group, a 3-fluorophenyl group, a 4-fluorophenyl group, a 2-bromophenyl group, a 3-bromophenyl group, a 4-bromophenyl group, a 2-iodophenyl group, a 4-iodophenyl group, a 4-methoxyphenyl group, a 3,4-dimethoxyphenyl group, a 3,4-dibenzoyloxyphenyl group, a 4-benzyloxyphenyl group, a 2-benzyloxyphenyl group, a 2-methoxyphenyl group, a 3-methoxyphenyl group and a 4-nitrophenyl group, etc.

[0030] In the present invention, R^2 in the compound (I) represented by the formula (I) represents a hydrogen atom, a C_1 to C_{10} alkyl group which may be substituted, a phenyl group which may be substituted or a C_2 to C_{10} alkenyl group which is not substituted.

[0031] "The C_1 to C_{10} alkyl group which may be substituted" represented by R^2 means (7) "a C_1 to C_{10} alkyl group having no substituent" or (8) "a C_1 to C_{10} alkyl group having a substituent(s)".

[0032] As "the C_1 to C_{10} alkyl group having no substituent" of (7) may be mentioned, for example, the above-mentioned straight or branched C_1 to C_{10} alkyl group.

[0033] As the substituent of (8) "the C_1 to C_{10} alkyl group having a substituent(s)", there may be mentioned, for example, a phthalimide group; a benzyloxy group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different, straight or branched C_1 to C_6 alkyl group portion; a halogen atom; an aryl group which may be substituted; a heteroaromatic ring group which may be substituted; a straight or branched C_1 to C_{10} alkoxy group; an acylamino group; a C_1 to C_{10} alkylthio group or a benzylthio group. Incidentally, the number and the position of the substituent are not limited.

[0034] "The aryl group which may be substituted" as a substituent for (8) "the C_1 to C_{10} alkyl group having a substituent(s)" means (8-1) "an aryl group having no substituent" or (8-2) "an aryl group having a substituent(s)".

[0035] As (8-1) "the aryl group having no substituent", there may be mentioned, for example, a phenyl group or a naphthyl group.

[0036] As the substituent for (8-2) "the aryl group having a substituent(s)", there may be mentioned, for example, a straight or branched C_1 to C_6 alkyl group; a nitro group; a benzyloxy group; a cyano group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different, straight or branched C_1 to C_6 alkyl group portion; a halogen atom; an acylamino group or a straight or branched C_1 to C_6 alkoxy group. Incidentally, the number and the position of the substituent are not limited.

[0037] "The heteroaromatic ring group which may be substituted" as a substituent for (8) "the C_1 to C_{10} alkyl group having a substituent(s)" means (8-3) "a heteroaromatic ring group having no substituent" or (8-4) "a heteroaromatic ring group having a substituent(s)".

[0038] As (8-3) "the heteroaromatic ring group having no substituent", there may be mentioned, for example, a furyl group, a thienyl group, an oxazolyl group, an isoxazolyl group, a thiazolyl group, an indolyl group or an imidazolyl group, etc.

[0039] The heteroaromatic ring group of (8-4) "the heteroaromatic ring group having a substituent(s)" has the same meaning as (8-3) "the heteroaromatic ring group having no substituent". As the substituent for (8-4) "the heteroaromatic ring group having a substituent(s)", there may be mentioned, for example, a straight or branched C_1 to C_{10} alkyl group; a halogen atom or an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion. Incidentally, the number and the position of the substituent are not limited.

[0040] In the present invention, "the phenyl group which may be substituted" represented by R^2 in the compound (I) represented by the formula (I) means a phenyl group or "a phenyl group having a substituent(s)". As the substituent

for "the phenyl group having a substituent(s)", there may be mentioned, for example, a benzyloxy group, a halogen atom, a straight or branched C₁ to C₆ alkoxy group; an acylamino group or a trialkylsilyloxy group having the same or different, straight or branched C₁ to C₆ alkyl group portion, etc. Incidentally, the number and the position of the substituent(s) are not limited.

[0041] In the present invention, "the C₂ to C₁₀ alkenyl group which is not substituted" represented by R² in the compound (I) represented by the formula (I) may be mentioned, for example, a straight or branched C₂ to C₁₀ alkenyl group.

[0042] Specific examples of such R² may include, for example, a hydrogen atom, a methyl group, an ethyl group, a propyl group, an isopropyl group, a butyl group, an isobutyl group, a sec-butyl group, a tert-butyl group, a pentyl group, a hexyl group, a heptyl group, an octyl group, a nonyl group, a decyl group, a 2-methoxycarbonylethyl group, a 3-methoxycarbonylpropyl group, a 4-methoxycarbonylbutyl group, a 1-trimethylsilyloxyethyl group, a benzyloxymethyl group, a 1-benzyloxyethyl group, a 1-benzyloxypropyl group, a 1-benzyloxybutyl group, a methoxymethyl group, a 1-tert-butoxyethyl group, a 1-ethoxyethyl group, a 1-hexyloxyethyl group, an isopropoxymethyl group, a n-propoxymethyl group, a 2-methylthioethyl group, a 2-ethylthioethyl group, a methylthiomethyl group, a butylthiomethyl group, a tert-butylthiomethyl group, a benzylthiomethyl group, a 2-trifluoroethyl group, a trifluoromethyl group, a 2-chloroethyl group, a fluoromethyl group, a 1-fluorobutyl group, a 1-fluoro-1-phenylmethyl group, a 1-fluoroethyl group, a 2-acetylaminomethyl group, a 3-benzoylaminopropyl group, a 4-formylaminobutyl group, a 4-acetylaminobutyl group, a 4-chloroacetaminobutyl group, a 4-phenylacetaminobutyl group, a 4-methoxycarbonylaminobutyl group, a 4-ethoxycarbonylaminobutyl group, a 4-allyloxycarbonylaminobutyl group, a 4-tertbutyloxycarbonylaminobutyl group, a 4-benzyloxycarbonylaminobutyl group, a 4-phthaloylaminobutyl group, a benzyl group, a 4-nitrobenzyl group, a 4-cyanobenzyl group, a 4-methoxycarbonylbenzyl group, a 4-trimethylsilyloxybenzyl group, a 4-benzyloxybenzyl group, a 3,4-dichlorobenzyl group, a 2-fluorobenzyl group, a 4-fluorobenzyl group, a 3,4-difluorobenzyl group, a 4-methylbenzyl group, a 2-methylbenzyl group, a 2,4-dimethylbenzyl group, a 4-isopropylbenzyl group, a 4-tert-butylbenzyl group, a 2-methoxybenzyl group, a 3-methoxybenzyl group, a 4-methoxybenzyl group, a 3,4-dimethoxybenzyl group, a 2,4-dimethoxybenzyl group, a 2-ethoxybenzyl group, a 4-isopropoxybenzyl group, a 4-tert-butoxybenzyl group, a 4-tert-butoxycarbonylaminobenzyl group, a 4-acetylaminobenzyl group, a 2-benzyloxycarbonylaminobenzyl group, a 1-phenylethyl group, a 1-(4-nitrophenyl)ethyl group, a 1-(4-bromophenyl)ethyl group, a 1-(4-fluorophenyl)ethyl group, a 1-(4-methoxyphenyl)ethyl group, a 1-(4-chlorophenyl)ethyl group, a 1-(1-naphthyl)ethyl group, a 1-(2-naphthyl)ethyl group, a diphenylmethyl group, a di(4-chlorophenyl)methyl group, a di(4-methoxyphenyl)methyl group, a trityl group, a 2-phenylethyl group, a 2-(4-benzyloxyphenyl)ethyl group, a furfuryl group, a thienylmethyl group, a thiazolylmethyl group, an isoxazolylmethyl group, an oxazolylmethyl group, a (4-N-methylimidazolyl)methyl group, an N-methylindolylmethyl group, a phenyl group, a 2-fluorophenyl group, a 4-benzyloxyphenyl group, a 4-methoxyphenyl group, a 2-chlorophenyl group, a 4-bromophenyl group, a 4-acetaminophenyl group, a 4-benzyloxycarbonylaminophenyl group, an ethynyl group, a 1-propenyl group, a 2-propenyl group, a 1-butenyl group, a 2-butenyl group, a 3-butenyl group, a 1-pentenyl group, a 2-pentenyl group, a hexenyl group, a heptenyl group, an octenyl group, a nonenyl group or a decenyl group.

[0043] In the present invention, R³ in the compound (I) represented by the formula (I) represents a C₁ to C₁₀ alkyl group which may be substituted, a C₃ to C₁₀ cycloalkyl group which may be substituted, a C₂ to C₁₀ alkenyl group which may be substituted (provided that a 2-alkenyl group is excluded) or a phenyl group which may be substituted.

[0044] "The C₁ to C₁₀ alkyl group which may be substituted" represented by R³ in the compound (I) represented by the formula (I) means (9) "a C₁ to C₁₀ alkyl group having no substituent" or (10) "a C₁ to C₁₀ alkyl group having a substituent(s)".

[0045] As (9) "the C₁ to C₁₀ alkyl group having no substituent", there may be mentioned, for example, the above-mentioned straight or branched C₁ to C₁₀ alkyl group.

[0046] As the substituent for (10) "the C₁ to C₁₀ alkyl group having a substituent(s)", there may be mentioned, for example, a benzyloxy group; an alkoxy carbonyl group having a straight or branched C₁ to C₆ alkyl group portion; an acylamino group; a halogen atom; "the aryl group which may be substituted" as "mentioned in (8-1) and (8-2)" or a C₁ to C₁₀ alkoxy group as "mentioned in (2-5)".

[0047] "The C₃ to C₁₀ cycloalkyl group which may be substituted" represented by R³ in the formula (I) represented by the formula (I) means (11) "a C₃ to C₁₀ cycloalkyl group having no substituent" or (12) "a C₃ to C₁₀ cycloalkyl group having a substituent(s)".

[0048] (11) "The C₃ to C₁₀ cycloalkyl group having no substituent" has the same meaning as "the C₃ to C₁₀ cycloalkyl group having no substituent" mentioned in (3), and (12) "the C₃ to C₁₀ cycloalkyl group having a substituent(s)" have the same meaning as "the C₃ to C₁₀ cycloalkyl group having a substituent(s) mentioned in (4).

[0049] "The C₂ to C₁₀ alkenyl group which may be substituted (provided that a 2-alkenyl group is excluded)" represented by R³ in the formula (I) represented by the formula (I) means (13) "a C₂ to C₁₀ alkenyl group having no substituent (provided that a 2-alkenyl group is excluded)" or (14) "a C₂ to C₁₀ alkenyl group having a substituent(s) (provided that a 2-alkenyl group is excluded)".

[0050] As (13) "the C₂ to C₁₀ alkenyl group having no substituent (provided that a 2-alkenyl group is excluded)", there may be mentioned, for example, a straight or branched C₂ to C₁₀ alkenyl group having no substituent (provided

that a 2-alkenyl group is excluded)".

[0051] Incidentally, when the process represented by the reaction formula (I) of the present invention as mentioned below is carried out by using a compound represented by the formula (II) containing a 2-alkenyl group in the molecule as R³, the compound represented by the formula (I) cannot be obtained.

5 [0052] As (14) "the C₂ to C₁₀ alkenyl group having a substituent(s) (provided that a 2-alkenyl group is excluded)", there may be mentioned, for example, a benzyloxy group; an alkoxycarbonyl group having a straight or branched C₁ to C₆ alkyl group portion; an acylamino group; a halogen atom; an aryl group which may be substituted "as mentioned in (8-1) and (8-2)" or a C₁ to C₁₀ alkoxy group "as mentioned in (2-5)". Incidentally, the number and the position of the substituent(s) are not limited.

10 [0053] "The phenyl group which may be substituted" represented by R³ in the compound (I) represented by the formula (I) means a phenyl group or "a phenyl group having a substituent(s)". As the substituent for "the phenyl group having a substituent(s)", there may be mentioned, for example, a straight or branched C₁ to C₆ alkyl group, a nitro group, and the above-mentioned halogen atom.

[0054] Specific examples of such R³ may include, for example, a methyl group, an ethyl group, a propyl group, an isopropyl group, a butyl group, an isobutyl group, a sec-butyl group, a tert-butyl group, a pentyl group, a hexyl group, a heptyl group, an octyl group, a nonyl group, a decyl group, a 2-methoxycarbonylethyl group, a 3-methoxycarbonylpropyl group, a 4-methoxycarbonylbutyl group, a 1-benzyloxyethyl group, a 3-benzyloxypropyl group, a 3-benzyloxyisobutyl group, a 2-methoxyethyl group, a 2-tert-butoxyethyl group, a 2-ethoxyethyl group, a 2-hexyloxyethyl group, a 2-isopropoxyethyl group, a 3-n-propoxypropyl group, a 2-trifluoroethyl group, a trifluoromethyl group, a 2-chloroethyl group, a fluoromethyl group, a 2-fluoroethyl group, a 1-fluorobutyl group, a 1-fluoro-1-phenylmethyl group, a 1-fluoroethyl group, a 2-bromoethyl group, a 2-acetylaminoethyl group, a 3-benzoylaminoethyl group, a 4-formylaminobutyl group, a 4-acetylaminoethyl group, a 4-chloroacetaminobutyl group, a 4-phenylacetaminobutyl group, a 4-methoxycarbonylaminobutyl group, a 4-ethoxycarbonylaminobutyl group, a 4-allyloxycarbonylaminobutyl group, a 4-tert-butyloxycarbonylaminobutyl group, a 4-benzyloxycarbonylaminobutyl group, a benzyl group, a 4-nitrobenzyl group, a 4-trimethylsilyloxybenzyl group, a 4-benzyloxybenzyl group, 3,4-dichlorobenzyl group, a 2-fluorobenzyl group, a 4-fluorobenzyl group, a 3,4-difluorobenzyl group, a 4-methylbenzyl group, a 2-methylbenzyl group, a 2,4-dimethylbenzyl group, a 4-isopropylbenzyl group, a 4-tert-butylbenzyl group, a 2-methoxybenzyl group, a 3-methoxybenzyl group, a 4-methoxybenzyl group, a 3,4-dimethoxybenzyl group, a 2,4-dimethoxybenzyl group, a 2-ethoxybenzyl group, a 4-isopropoxybenzyl group, a 4-tert-butoxybenzyl group, a 4-tert-butoxycarbonylaminobenzyl group, a 4-acetylaminoethyl group, a 2-benzyloxycarbonylbenzyl group, a 1-phenylethyl group, a 1-(4-nitrophenyl)ethyl group, a 1-(4-bromophenyl)ethyl group, a 1-(4-fluorophenyl)ethyl group, a 1-(4-methoxyphenyl)ethyl group, a 1-(4-chlorophenyl)ethyl group, a 1-(1-naphthyl)ethyl group, a 1-(2-naphthyl)ethyl group, a diphenylmethyl group, a di(4-chlorophenyl)methyl group, a di(4-methoxyphenyl)methyl group, a trityl group, a 2-phenylethyl group, a 2-(4-benzyloxyphenyl)ethyl group, an ethenyl group, a 1-propenyl group, a 1-butenyl group, a 3-butenyl group, a 1-pentenyl group, a 3-pentenyl group, a 4-pentenyl group, a 5-hexenyl group, a 6-heptenyl group, a 7-octenyl group, a 8-nonyl group, a 9-decenyl group, a 3-methoxycarbonyl-1-propenyl group, a 1-methoxycarbonylethynyl group, a 2-benzyloxymethylethynyl group, a 3-chloro-4-pentenyl group, a 4-chloro-3-butenyl group, a 4-phenyl-3-butenyl group, a 5-phenyl-4-pentenyl group, a 5-benzyloxy-3-butenyl group, a 6-methoxy-3-hexenyl group, a cyclopropyl group, a cyclobutyl group, a cyclopentyl group, a cyclohexyl group, a cycloheptyl group, a cyclooctyl group, a cyclononyl group, a cyclodecyl group, a 1-methoxycarbonylcyclopropyl group, a 2-phenylcyclopropyl group, a 2-fluorocyclopropyl group, a 2-chlorocyclopropyl group, a 1-benzyloxymethylcyclopentyl group, a 3-cyanocyclopentyl group, a 2-acetaminocyclohexyl group, a 2-benzoylaminoethyl group, a 2-methoxycarbonylaminocyclohexyl group, a 2-tert-butoxycarbonylaminocyclohexyl group, a 2-benzyloxycarbonylaminocyclohexyl group, a 2-methoxycyclohexyl group, a 2-chlorocyclohexyl group, a 2-norbornyl group, a bornyl group, a 2-adamantyl group, an N-benzosulfonyl-N-(3,5-dimethylphenyl)-aminobornyl group, a 4-methylcyclohexyl group, a 2-methylcyclohexyl group, a 2,3-dimethylcyclohexyl group, a 1-methoxycarbonylcyclohexyl group, a 2-trimethylsilyloxycyclohexyl group, a 2-benzyloxycyclohexyl group, a 4-benzyloxycyclohexyl group, a 4-tert-butylcyclohexyl group, a menthyl group, a 8-phenylmenthyl group, a phenyl group, a 2-methylphenyl group, a 4-methylphenyl group, a 2,5-di-tert-butyl-4-methyl group, a 4-tert-butyl group, a 2-chlorophenyl group, a 4-chlorophenyl group, a 4-fluorophenyl group, a 2-fluorophenyl group, a 3-chlorophenyl group or a 4-nitrophenyl group, etc.

50 [0055] As the specific compounds of the compound (I) represented by the formula (I), there may be mentioned, for example, 3-benzyl-5-methoxy-2(3H)oxazolone, 3-benzyl-4-methyl-5-methoxy-2(3H)oxazolone, 3-(4-nitrobenzyl)-5-ethoxy-2(3H)oxazolone, 3-benzyl-5-(*l*)-menthyl-5-methoxy-2(3H)oxazolone, 3-(4-methylbenzyl)-4-methyl-5-methoxy-2(3H)oxazolone, 3-(1-phenylethyl)-5-methoxy-2(3H)oxazolone, 3-((*S*)-1-phenylethyl)-5-isopropoxy-2(3H)oxazolone, 3-((*R*)-1-phenylethyl)-5-methoxy-2(3H)oxazolone, 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, 3-diphenylmethyl-4-methyl-5-methoxy-2(3H)oxazolone, 3-diphenylmethyl-5-(*l*)-menthyl-5-methoxy-2(3H)oxazolone, 3-diphenylmethyl-5-((1*S*,2*S*,5*R*))-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone, 3-benzyl-5-((1*S*,2*S*,5*R*))-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone, 3-diphenylmethyl-4-isopropyl-5-isopropoxy-2(3H)oxazolone, 3-diphenylmethyl-4-phenylmethyl-5-methoxy-2(3H)oxazolone, 3-diphenylmethyl-4-isobutyl-5-ethoxy-2(3H)oxazolone,

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benzoyl group.

[0060] "The C_1 to C_{20} alkyl group which may be substituted" represented by R^4 means (15) "a C_1 to C_{20} alkyl group having no substituent" or (16) "a C_1 to C_{20} alkyl group having a substituent(s)".

[0061] As (15) "the C_1 to C_{20} alkyl group having no substituent", there may be mentioned, for example, a straight or branched C_1 to C_{20} alkyl group such as a methyl group, an ethyl group, a propyl group (including an isomer), a butyl group (including isomers thereof), a pentyl group (including isomers thereof), a hexyl group (including isomers thereof), a heptyl group (including isomers thereof), an octyl group (including isomers thereof), a nonyl group (including isomers thereof), a decyl group (including isomers thereof), an undecyl group (including isomers thereof), a dodecyl group (including isomers thereof), a tridecyl group (including isomers thereof), a tetradecyl group (including isomers thereof), a pentadecyl group (including isomers thereof), a hexadecyl group (including isomers thereof), a heptadecyl group (including isomers thereof), an octadecyl group (including isomers thereof), a nonadecyl group (including isomers thereof) or an eicosyl group (including isomers thereof), etc.

[0062] As the substituent for (16) "the C_1 to C_{20} alkyl group having a substituent(s)", there may be mentioned, for example, a cyano group; a benzyloxy group; a phthalimino group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different straight or branched C_1 to C_6 alkyl group portion; an acylamino group; a halogen atom; an aryl group as "described in (2-1) and (2-2)"; a heteroaromatic ring group as "described in (8-3) and (8-4)" or a C_1 to C_{10} alkoxy group as "described in (2-5) and (2-6)".

[0063] "The C_2 to C_{20} alkenyl group which may be substituted" represented by R^4 in the compound (II) represented by the formula (II) has the same meaning as (17) "a C_2 to C_{20} alkenyl group having no substituent" or (18) "a C_2 to C_{20} alkenyl group having a substituent(s)".

[0064] As (17) "the C_2 to C_{20} alkenyl group having no substituent", there may be mentioned, for example, a straight or branched C_2 to C_{10} alkenyl group.

[0065] As the substituent for (18) "the C_2 to C_{20} alkenyl group having a substituent(s)", there may be mentioned, for example, a cyano group; a benzyloxy group; an acylamino group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different straight or branched C_1 to C_6 alkyl group portion; a halogen atom; an aryl group as "described in (2-3) and (2-4)"; a C_3 to C_{10} cycloalkyl group or a heteroaromatic ring group.

[0066] "The C_3 to C_{10} cycloalkyl group which may be substituted" represented by R^4 in the compound (II) represented by the formula (II) has the same meaning as (19) "a C_3 to C_{10} cycloalkyl group having no substituent" or (20) "a C_3 to C_{10} cycloalkyl group having a substituent(s)".

[0067] As (19) "the C_3 to C_{10} cycloalkyl group having no substituent", there may be mentioned, for example, a cyclopropyl group, a cyclobutyl group, a cyclopentyl group, a cyclohexyl group, a cycloheptyl group, a cyclooctyl group, a cyclononyl group or a cyclodecyl group.

[0068] As the substituent for (20) "the C_3 to C_{10} cycloalkyl group having a substituent(s)", there may be mentioned, for example, a cyano group; a benzyloxy group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different straight or branched C_1 to C_6 alkyl group portion; an acylamino group; a halogen atom; an aryl group or a C_1 to C_6 alkoxy group. Incidentally, the number and the position of the substituent(s) are not limited.

[0069] "The C_2 to C_{20} alkynyl group which may be substituted" represented by R^4 in the compound (II) represented by the formula (II) has the same meaning as (21) "a C_2 to C_{20} alkynyl group having no substituent" or (22) "a C_2 to C_{20} alkynyl group having a substituent(s)".

[0070] As (21) "the C_2 to C_{20} alkynyl group having no substituent", there may be mentioned, for example, a straight or branched C_2 to C_{20} alkynyl group.

[0071] As the substituent for (22) "the C_2 to C_{20} alkynyl group having a substituent(s)", there may be mentioned, for example, a cyano group; a benzyloxy group; an acylamino group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different straight or branched C_1 to C_6 alkyl group portion; a halogen atom or an aryl group.

[0072] "The C_6 to C_{20} aryl group which may be substituted" represented by R^4 in the compound (II) represented by the formula (II) has the same meaning as (23) "a C_6 to C_{20} aryl group having no substituent" or (24) "a C_6 to C_{20} aryl group having a substituent(s)".

[0073] As (23) "the C_6 to C_{20} aryl group having no substituent", there may be mentioned, for example, a phenyl group, a naphthyl group, an anthranyl group, a phenanthryl group, etc.

[0074] The aryl group of (24) "the C_6 to C_{20} aryl group having a substituent(s)" has the same meaning as (23) "the C_6 to C_{20} aryl group having no substituent".

[0075] As the substituent for (24) "the C_6 to C_{20} aryl group having a substituent(s)", there may be mentioned, for example, a nitro group; a cyano group; a benzyloxy group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different straight or branched C_1 to C_6 alkyl group portion; a halogen atom; a straight or branched C_1 to C_6 alkyl group or a straight or branched C_1 to C_6 alkoxy group.

[0076] "The 5- or 6-membered heteroaromatic ring group containing 1 or 2 hetero atoms selected from N, O and S, which may be substituted" represented by R^4 in the compound (II) represented by the formula (II) has the same meaning as (25) "a 5- or 6-membered heteroaromatic ring group containing 1 or 2 hetero atoms selected from N, O and S having no substituent" or (26) "a 5- or 6-membered heteroaromatic ring group containing 1 or 2 hetero atoms selected from N, O and S having a substituent(s)".

[0077] As the heteroaromatic ring group of (25) "the 5- or 6-membered heteroaromatic ring group containing 1 or 2 hetero atoms selected from N, O and S having no substituent", there may be mentioned, for example, a furyl group, a thienyl group, a pyrrolyl group, a 2H-pyrrolyl group, an oxazolyl group, an isoxazolyl group, a thiazolyl group, an isothiazolyl group, an imidazolyl group, a pyrazolyl group, a pyranlyl group, a pyridyl group, a pyridazyl group, a pyrimidyl group or a pyrazinyl group, etc.

[0078] As the substituent for (26) "the 5- or 6-membered heteroaromatic ring group containing 1 or 2 hetero atoms selected from N, O and S having a substituent(s)", there may be mentioned, for example, a nitro group; a cyano group; a benzyloxy group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different, straight or branched C_1 to C_6 alkyl group portion; a halogen atom; a straight or branched C_1 to C_6 alkyl group; or a straight or branched C_1 to C_6 alkoxy group. Incidentally, the number and the position of the substituent are not limited.

[0079] "The C_1 to C_6 alkoxycarbonyl group which may be substituted" represented by R^4 in the compound (II) represented by the formula (II) means (27) "a C_1 to C_6 alkoxycarbonyl group having no substituent" or (28) "a C_1 to C_6 alkoxycarbonyl group having a substituent(s)".

[0080] As the alkoxycarbonyl group of (27) "the C_1 to C_6 alkoxycarbonyl group having no substituent", there may be mentioned, for example, a straight or branched C_1 to C_6 alkoxycarbonyl group such as a methoxycarbonyl group, an ethoxycarbonyl group, a n-propionyloxycarbonyl group, an isopropionyloxycarbonyl group, a n-butoxycarbonyl group, a tert-butoxycarbonyl group, a n-pentyloxycarbonyl group, a n-hexyloxycarbonyl group, etc.

[0081] As the substituent for (28) "the C_1 to C_6 alkoxycarbonyl group having a substituent(s)", there may be mentioned, for example, a nitro group; a cyano group; a benzyloxy group; an alkoxycarbonyl group having a straight or branched C_1 to C_6 alkyl group portion; a trialkylsilyloxy group having the same or different straight or branched C_1 to C_6 alkyl group portion; a halogen atom; a straight or branched C_1 to C_6 alkoxy group, etc. Incidentally, the number and the position of the substituent(s) are not limited.

[0082] Specific examples of such R^4 may include, for example, a hydrogen atom, a methyl group, an ethyl group, a n-propyl group, an isopropyl group, a n-butyl group, an isobutyl group, a sec-butyl group, a tert-butyl group, a n-pentyl group, a n-hexyl group, a n-heptyl group, a n-octyl group, a n-nonyl group, a n-decyl group, a n-undecyl group, a n-dodecyl group, a n-tridecyl group, a n-tetradecyl group, a n-pentadecyl group, a n-hexadecyl group, a n-heptadecyl group, a n-octadecyl group, a n-nonadecyl group, an n-eicosyl group, a methoxymethyl group, an acetyloxymethyl group, a benzyloxymethyl group, a tert-butyloxymethyl group, a 1-methoxyethyl group, a 1-benzyloxyethyl group, a 1-benzyloxypropyl group, a 1-benzyloxybutyl group, a 1-phenylethyl group, a 2-phenylethyl group, a 3-phenylpropyl group, a phenylmethyl group, a 2-chloroethyl group, a 2-fluoroethyl group, a monofluoromethyl group, a trifluoromethyl group, a BocNH-methyl group (Boc = tert-butoxycarbonyl), a CbzNH-methyl group (Cbz = benzyloxycarbonyl), a benzoyl group, an acetyl group, a methoxycarbonyl group, an ethoxycarbonyl group, a menthyloxycarbonyl group, a methylthiomethyl group, a methoxycarbonylmethyl group, a 3-BocNH-propyl group, a 4-BocNH-butyl group, an acylaminoethyl group, a 2-benzyloxypropyl group, a cyclohexylmethyl group, a 1-methyl-3-methoxycarbonylbutyl group, a 1-cyclohexylethyl group, a 2-(N-Boc-piperidyl)methyl group, a vinyl group, a 1-propenyl group, a 2-propenyl group, a 1-butenyl group, a 2-butenyl group, a styryl group, a p-nitrostyryl group, a p-chlorostyryl group, a p-methylstyryl group, a p-methoxystyryl group, a 3-phenyl-2-propenyl group, a 3-phenyl-1-propenyl group, a 2-hexenyl group, a 2-heptenyl group, a 2-pentenyl group, a 2,4-hexadienyl group, a 1-methyl-3-pentenyl group, a 1-tetradecynyl group, a 1-pentadecynyl group, a 1-hexadecenyl group, a 2,3,12-tribenzyloxy-3-heptadecynyl group, a 2,3,12-tri-tert-butyltrimethylsilyloxy-3-heptadecynyl group, a 2-furyl-vinyl group, a 2-cyclohexyl-vinyl group, a phenylethyl group, a 3-phenyl-1-propargyl group, a propargyl group, a 1-butyne group, a 1-pentyne group, a 1-hexyne group, a 2-heptyne group, a 1-octyne group, a 1-nonyne group, a cyclopropyl group, a cyclohexyl group, a cyclobutyl group, a cyclopentyl group, a 2-phenylcyclopropyl group, a 2,2-dimethoxycarbonylcyclopropyl group, a phenyl group, a o-fluorophenyl group, a m-fluorophenyl group, a p-fluorophenyl group, a o-chlorophenyl group, a m-chlorophenyl group, a p-chlorophenyl group, a p-bromophenyl group, a p-methylphenyl group, a p-methoxyphenyl group, a o-methoxyphenyl group, a m-methoxyphenyl group, a p-benzyloxyphenyl group, a m-benzyloxyphenyl group, a o-benzyloxyphenyl group, a 3,4-dibenzyloxyphenyl group, a 3,4-methylenedioxyphenyl group, a 3,4-dimethoxyphenyl group, a p-nitrophenyl group, a 3,5-dibromo-4-benzyloxyphenyl group, a 3,5-dichloro-4-benzyloxyphenyl group, a p-tert-butyltrimethylsilyloxyphenyl group, a furyl group, a thienyl group, a thiazolyl group, an isoxazolyl group, an oxazolyl group, a 4-N-methylimidazolyl group, a N-methylindolyl group, etc.

[0083] As the compound (III), compounds comprising a combination of the substituents described as R^1 , R^2 , R^3 and R^4 as mentioned above may be mentioned. As the specific compound (III) thereof, the following compounds may be mentioned.

[0084] There may be mentioned, for example, 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(3,4-dimethoxyphenyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(4-benzyloxyphenyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(3,4-dibenzoyloxyphenyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(3,4-ditrimethylsilyloxyphenyl)-2-oxazolidinone, 3-diphenylmethyl-4-ethoxycarbonyl-5-phenyl-2-oxazolidinone, 3-diphenylmethyl-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone, 3-diphenylmethyl-4-tert-butoxycarbonyl-5-phenyl-2-oxazolidinone, 3-diphenylmethyl-4-((*l*)-menthyloxy)carbonyl-5-phenyl-2-oxazolidinone, 3-diphenylmethyl-4-(8-phenylmenthyloxy)carbonyl-5-phenyl-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-methyl-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(3-BocNH-propyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(methoxycarbonylmethyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-tridecyl-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(1-methyl-3-pentenyl)-2-oxazolidinone, 3-methyl-4-methoxycarbonyl-5-(1-methyl-3-pentenyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-(1-pentadecenyl)-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-hydroxymethyl-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-benzyloxymethyl-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-acetyloxymethyl-2-oxazolidinone, 3-diphenylmethyl-4-methoxycarbonyl-5-trimethylsilyloxymethyl-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-(4-benzyloxyphenyl)-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-(3,4-dimethoxyphenyl)-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-methyl-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-benzyloxymethyl-2-oxazolidinone, 3-(1-phenylethyl)-4-methoxycarbonyl-5-methoxycarbonylmethyl-2-oxazolidinone, 3-(1-phenylethyl)-4-((*l*)-menthyloxy)carbonyl-5-phenyl-2-oxazolidinone, 3-(1-phenylethyl)-4-((*l*)-menthyloxy)carbonyl-5-isopropyl-2-oxazolidinone, 3-(1-phenylethyl)-4-((*l*)-menthyloxy)carbonyl-5-(2-phenylethyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-((*l*)-menthyloxy)carbonyl-5-phenyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-dimethoxyphenyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-isopropoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-isopropoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-isopropoxycarbonyl-5-(2-phenylethynyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-isopropoxycarbonyl-5-methyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-isopropoxycarbonyl-5-cyclopropyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-cyclohexyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-isopropoxycarbonyl-5-(1-propenyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(4-benzyloxyphenyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-dibenzoyloxyphenyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-ditrimethylsilyloxyphenyl)-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-ethoxycarbonyl-5-phenyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-tert-butoxycarbonyl-5-phenyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-((*l*)-menthyloxy)carbonyl-5-phenyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-(8-phenylmenthyloxy)carbonyl-5-phenyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone, 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone, 3-benzyl-4-(8-phenylmenthyloxy)carbonyl-5-phenyl-2-oxazolidinone, 3-benzyl-4-(8-phenylmenthyloxy)carbonyl-5-isopropyl-2-oxazolidinone, 3-benzyl-4-(8-phenylmenthyloxy)carbonyl-5-(2-phenylethyl)-2-oxazolidinone, 3-phenyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone, 3-(*o*-chlorophenyl)-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone, 3-phenyl-4-methoxycarbonyl-4-methyl-5-methyl-2-oxazolidinone, 3-(*p*-chlorophenyl)-4-methoxycarbonyl-4-ethyl-5-phenyl-2-oxazolidinone, 3-phenyl-4-isopropoxyloxy)carbonyl-4-methyl-5-phenyl-2-oxazolidinone, 3-diphenylmethyl-4-phenoxy)carbonyl-4-methyl-5-phenyl-2-oxazolidinone, 3-(1-phenylethyl)-4-phenoxy)carbonyl-5-phenyl-2-oxazolidinone, 3-phenyl-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone, etc.

[0085] As the Lewis acid catalyst to be used in the present invention, the following organometallic compounds, halides, or trifluoromethane sulfonates of elements from Group 2 (Group IIa) to Group 14 (Group IVa) (provided that car-

bon is excluded) of the Periodic Table, halides or trifluoromethane sulfonates of a Lanthanoid group metal can be specifically used.

[0086] Specifically, there may be mentioned the compound represented by the following formula (IV):



wherein R^5 represents a C_1 to C_{10} alkyl group or a C_6 to C_{20} aryl group; X represents a halogen atom; M represents Al, B, Sn or Ti; m and n each represents a number of 0, 1, 2, 3 or 4; provided that $m + n$ is 2, 3 or 4.

[0087] The alkyl group represented by R^5 in the compound (IV) represented by the formula (IV) represents a C_1 to C_{10} alkyl group, and there may be mentioned, for example, a straight or branched C_1 to C_{10} alkyl group such as a methyl group, an ethyl group, a propyl group (including an isomer), a butyl group (including isomers thereof), a pentyl group (including isomers thereof), a hexyl group (including isomers thereof), a heptyl group (including isomers thereof), an octyl group (including isomers thereof), a nonyl group (including isomers thereof) or a decyl group (including isomers thereof), etc., preferably a C_1 to C_6 alkyl group, more preferably a methyl group, an ethyl group, a propyl group, an isopropyl group, a butyl group, an isobutyl group, a sec-butyl group or a tert-butyl group.

[0088] The aryl group represented by R^5 in the compound (IV) represented by the formula (IV) has the same meaning as the aryl group as "described in (23) and (24)".

[0089] X in the compound (IV) represented by the formula (IV) represents, for example, a halogen atom such as a fluorine atom, a chlorine atom, a bromine atom, an iodine atom, etc. M represents Al, B, Sn or Ti. m and n each represents a number of 0, 1, 2, 3 or 4. However, $m + n$ is 2, 3 or 4.

[0090] As the specific compounds of the compound (IV) represented by the formula (IV), there may be mentioned, for example, trialkyl aluminum such as trimethyl aluminum, triethyl aluminum, tripropyl aluminum, triisopropyl aluminum, tri-n-butyl aluminum, triisobutyl aluminum, tri-sec-butyl aluminum, tri-tert-butyl aluminum, triphenyl aluminum, trihexyl aluminum, trioctyl aluminum, tridecyl aluminum, etc.; dialkyl aluminum halide such as dimethyl aluminum chloride, diethyl aluminum chloride, diethyl aluminum bromide, diisobutyl aluminum chloride, etc.; dialkyl aluminum hydride such as diethyl aluminum hydride, diisobutyl aluminum hydride, etc.; an alkyl aluminum dihalide such as methyl aluminum dichloride, ethyl aluminum dichloride, isobutyl aluminum dichloride, ethyl aluminum dibromide, etc.; aluminum halides such as aluminum (III) chloride, aluminum (III) bromide, aluminum (III) iodide, aluminum (III) fluoride, etc.; boron trihalides such as boron trifluoride, boron trichloride, boron tribromide, etc.; trihaloboranes such as trifluoroborane, trichloroborane, tribromoborane, etc.; triarylboranes such as triphenylborane, tri(4-fluorophenyl)borane, tris(pentafluorophenyl)borane, etc.; arylborane dihalides such as phenyl dichloroborane, 4-chlorophenyl dichloroborane, phenyl dibromoborane, etc.; arylborane halides such as dimethylfluoroborane, diphenylfluoroborane, etc.; trialkyl boranes such as triethyl borane, tributyl borane, etc.; tin halides such as tin (IV) chloride, tin (II) chloride, tin (IV) bromide, tin (II) bromide, etc.; tetraalkyl tins such as tetraethyl tin, tetrabutyl tin, tetraisopropyl tin, etc.; tetraaryl tins such as tetraphenyl tin, etc.; alkylaryl tins such as benzyltriphenyl tin, pentafluorophenyltrimethyl tin, etc.; trialkyl tin halides such as trimethyl tin bromide, trimethyl tin chloride, triethyl tin bromide, tributyl tin chloride, tribenzyl tin chloride, etc.; triaryl tin halides such as triphenyl tin chloride, etc.; trialkyl tin hydrides such as tributyl tin hydride, etc.; titanium halides such as titanium (IV) chloride, titanium (IV) bromide, titanium (IV) iodide, titanium (IV) fluoride, etc.

[0091] Also, as a Lewis acid, the compound represented by the formula (V):



wherein R^6 represents a C_1 to C_{10} alkyl group or a C_6 to C_{20} aryl group; X represents a halogen atom; M represents Al, B, Sn or Ti; m' and n' each represents a number of 0, 1, 2, 3 or 4; provided that $m' + n'$ is 3 or 4,

can be also mentioned.

[0092] The C_1 to C_{10} alkyl group or the C_6 to C_{20} aryl group represented by R^6 in the compound (V) represented by the formula (V) have the same meanings as the C_1 to C_{10} alkyl group or the C_6 to C_{20} aryl group represented by R^5 , respectively.

[0093] X in the compound (V) represented by the formula (V) represents, for example, a halogen atom such as a fluorine atom, a chlorine atom, a bromine atom, an iodine atom, etc. M represents Al, B, Sn or Ti. m' and n' each represents a number of 0, 1, 2, 3 or 4. However, $m' + n'$ is 3 or 4.

[0094] As the specific compounds of the compound (V) represented by the formula (IV), there may be mentioned, for example, alkoxytitanium trihalides such as trichloromethoxy titanium, trichloroethoxy titanium, trichloroisopropoxy titanium, trichloro-n-butoxy titanium, tribromoethoxy titanium, tribromoisobutoxy titanium, etc.; dialkoxy titanium dihalides such as dichlorodimethoxy titanium, dichlorodiethoxy titanium, dichlorodi-n-butoxy titanium, dichlorodiisopropoxy titanium, dibromodiethoxy titanium, etc.; trialkoxy titanium halides such as chlorotrimethoxy titanium, chlorotriethoxy

titanium, chlorotri-n-butoxy titanium, bromotriethoxy titanium, etc.; titanium (IV) alkoxides such as titanium (IV) methoxide, titanium (IV) ethoxide, titanium (IV) isopropoxide, titanium (IV) isobutoxide, titanium (IV) n-butoxide, etc.; dialkyl aluminum alkoxides such as dimethyl aluminum methoxide, diethyl aluminum ethoxide, diethyl aluminum isopropoxide, dibutyl aluminum butoxide, etc.; dialkyl aluminum aryloxides such as diethyl aluminum phenoxide, diethyl aluminum (4-fluorophenoxide), etc.; aluminum (III) alkoxides such as aluminum (III) methoxide, aluminum (III) isopropoxide, aluminum (III) butoxide, etc.; aluminum (III) aryloxides such as aluminum (III) phenoxide, etc.; triaryloxy boranes such as triphenoxy borane, etc.; trialkoxy boranes such as trimethoxy borane, tributoxy borane, etc.; diaryloxyborane dihalides such as chlorodiphenoxy borane, bromodiphenoxy borane, fluorodiphenoxy borane, etc.; diaryloxy borane dihalides such as dichlorophenoxo borane, etc.

[0095] Moreover, as the Lewis acid, a compound represented by the formula (VI):



Wherein R^7 , R^8 and R^9 each independently represents a C_1 to C_{10} alkyl group or a C_6 to C_{20} aryl group; X' represents a halogen atom or $-OSO_2CF_3$.

[0096] The C_1 to C_{10} alkyl group or the C_6 to C_{20} aryl group represented by R^7 , R^8 and R^9 in the compound (VI) represented by the formula (VI) have the same meanings as the C_1 to C_{10} alkyl group or the C_6 to C_{20} aryl group represented by R^5 , respectively.

[0097] X' represents, for example a halogen atom such as a fluorine atom, a chlorine atom, a bromine atom, an iodine atom, etc., or a $-OSO_2CF_3$ group.

[0098] Specific compound represented of the compound (VI) represented by the formula (VI) may include, for example, trialkylsilyl halides such as trimethylsilyl chloride, trimethylsilyl bromide, trimethylsilyl iodide, triethylsilyl chloride, triethylsilyl bromide, triethylsilyl iodide, tert-butyldimethylsilyl chloride, tert-butyldimethylsilyl bromide, tert-butyldimethylsilyl iodide, triisopropylsilyl chloride, triisopropylsilyl bromide, triisopropylsilyl iodide, etc.; trialkylsilyl triflates such as trimethylsilyl triflate, triethylsilyl triflate, tert-butyldimethylsilyl triflate, triisopropylsilyl triflate, etc.; triarylsilyl halides such as triphenylsilyl chloride, triphenylsilyl bromide, triphenylsilyl iodide, etc.; triarylsilyl triflates such as triphenylsilyl triflate, etc.; alkylarylsilyl halides such as tert-butyldiphenylsilyl chloride, tert-butyldiphenylsilyl bromide, tert-butyldiphenylsilyl iodide, methyldiphenylsilyl chloride, methyldiphenylsilyl bromide, methyldiphenylsilyl iodide, etc.; alkylarylsilyl triflates such as tert-butyldiphenylsilyl triflate, methyldiphenylsilyl triflate, etc.

[0099] In addition, there may be mentioned metal halides or triflates such as zirconium (IV) chloride, zinc (II) chloride, zinc (II) bromide, iron (III) chloride, iron (II) chloride, iron (III) bromide, iron (II) bromide, iron (III) iodide, iron (II) iodide, magnesium chloride, magnesium bromide, tin triflate (II), etc.; halides or triflates of rare earth metals such as lanthanum (III) triflate, lanthanum (III) chloride, praseodymium (III) triflate, neodymium (III) triflate, samarium (III) triflate, samarium (II) iodide, samarium (III) chloride, europium (III) triflate, gadolinium (III) triflate, dysprosium (III) triflate, holmium (III) triflate, erbium (III) triflate, ytterbium (III) triflate, lutetium (III) triflate, scandium (III) triflate, cerium (III) chloride, etc.

[0100] The 4-alkoxycarbonyl-2-oxazolidinone compound represented by the compound (III) can be produced according to the method of the reaction formula (1) as mentioned above.

[0101] That is, the production of the compound (III) can be accomplished by reacting the 5-alkoxy-2(3H)oxazolone compound represented by the formula (I) with the aldehyde compound represented by the formula (II) in the presence of a Lewis acid catalyst in a solvent.

[0102] An amount of the Lewis acid catalyst to be used may be mentioned in a ratio of usually 0.001 to 5.0 equivalents, preferably 0.005 to 0.5 equivalent based on 1 mol of the compound (I).

[0103] As the solvent to be used, it is not particularly limited so long as it directly participate the present reaction, and there may be mentioned, for example, aromatic hydrocarbons such as benzene, toluene, xylene, etc., chlorinated hydrocarbons such as chlorobenzene, dichlorobenzene, methylene chloride, chloroform, carbon tetrachloride, 1,2-dichloroethane, etc., esters such as methyl acetate, ethyl acetate, butyl acetate, etc., ethers such as diethyl ether, diisopropyl ether, dibutyl ether, dimethoxyethane, tetrahydrofuran, dioxane, etc., dimethylformamide, dimethylsulfoxide, acetonitrile, propionitrile, water, etc., but preferably toluene, xylene, methylene chloride, 1,2-dichloroethane, acetonitrile, propionitrile are preferably used.

[0104] An amount of the solvent can be used in an amount of 100 to 5000 ml, preferably 300 to 3000 ml based on 1 mol of the compound (I).

[0105] Synthetic step of the compound (III) is preferably in an inert gas atmosphere such as nitrogen, argon, helium, etc., the reaction temperature is -100 to 100°C , preferably -80 to 40°C , and the reaction time can be suitably selected depending on the reaction time, concentrations of the starting materials to be charged, kinds of the starting materials to be charged, etc., but generally 1 to 10 hours. An amount of the starting material is that the compound (II) is used in an amount of 0.8 to 3-fold moles, preferably 0.9 to 1.2-fold moles based on the compound (I).

[0106] As a method of obtaining a reaction mixture containing the formed compound (III) in the present invention, usual washing operation and separating operation are carried out in combination. For example, a formed salt is removed by filtration operation, the filtrate is subjected to removal operation such as washing with water, dehydration by a drier and concentration of an organic solvent whereby a crude product of the compound (III) can be obtained. When the compound is to be further purified, purification can be carried out by the conventionally known means such as column chromatography, recrystallization, etc.

[0107] Also, the compound (III) of the present invention can be obtained by applying the specific preparation method described in Examples.

[0108] Incidentally, in the resulting compound (III), β -hydroxy- α -amino acid can be easily derived by subjecting to hydrolysis at the ester portion or carbamate portion, as reported in, for example, J. Am. Chem. Soc., 1996, 118, 3584-3590 and it would be clear. Moreover, when R¹ is a diphenylmethyl group, a benzyl group, a 1-phenylethyl group, a 1-(1-naphthyl)ethyl group, deprotection is possible by hydrogenation reaction using a Pd catalyst to give a free amino group which is also described in the above-mentioned reference and would be clear.

Examples

[0109] In the following, the present invention is specifically explained by referring to Examples. Incidentally, the scope of the present invention is not limited by these Examples.

Example 1: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0110] In 4 ml of methylene chloride were dissolved 0.281 g (1.0 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone and 0.106 g (1.0 mmol) of benzaldehyde, and the mixture was cooled to -78°C under argon atmosphere, and 14 mg (0.1 mmol) of BF₃ · Et₂O was added to the mixture and the resulting mixture was reacted at -78°C for 3 hours under stirring.

[0111] Moreover, the temperature was raised to -20°C and 15 ml of an aqueous saturated sodium hydrogen carbonate solution was added to the resulting reaction mixture, and the resulting mixture was extracted with 10 ml of methylene chloride. The organic layer was washed with a saturated saline solution and dried over anhydrous magnesium sulfate, and concentrated under reduced pressure to obtain 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-oxazolidinone as white crystal (0.4 g). According to the HPLC analysis, a formation ratio of cis/trans isomers of the formed product was 2/98. Yield based on 3-diphenylmethyl-5-methoxy-2(3H)oxazolone was substantially quantitative.

Melting point: 99 to 102°C

IR (KBr, cm⁻¹): 1760 (s).

¹H-NMR (δ , CDCl₃) (trans-isomer): 3.40 (s, 3H), 4.21 (d, J=3.9Hz), 5.44 (d, J=3.9Hz, 1H), 6.27 (s, 1H), 7.0-7.45 (m, 15H).

MS (CI, i-C₄H₁₀) m/z: 388 (MH⁺).

Example 2: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0112] Under argon atmosphere, 141 mg (0.5 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone was dissolved in 3 ml of methylene chloride, and after cooling the mixture to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-t-butyltrimethylsilyl triflate were added to the mixture and reacted under stirring for one hour.

[0113] Moreover, after elevating the temperature to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 174 mg (0.448 mmol) of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone as white crystal. (Yield based on 3-diphenylmethyl-5-methoxy-2(3H)oxazolone: 90%.) According to HPLC analysis, a formation ratio of cis/trans-isomers of the formed product was 6/94.

Example 3: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0114] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 141 mg (0.5 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-SnCl₄ were added to the solution and the mixture was reacted under stirring for 2 hours.

[0115] Moreover, the temperature of the mixture was raised to room temperature (20°C), and after stirring for fur-

ther 15 hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrated residue was quantitated by the HPLC method, yield of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 43%. In this case, a formation ratio of cis/trans-isomers of the formed product was 7/93.

Example 4: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0116] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 141 mg (0.5 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-TiBr₄ were added to the solution and the mixture was reacted under stirring for 2 hours. Then, the temperature of the mixture was raised to room temperature, and after stirring for further 16 hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrated residue was quantitated by the HPLC method, yield of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 26%. A formation ratio of cis/trans-isomers of the formed product was 8/92.

Example 5: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0117] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 141 mg (0.5 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour. After the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 184 mg (0.476 mmol) of 3-diphenylmethyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone as white crystal. (Yield based on 3-diphenylmethyl-5-methoxy-2(3H)oxazolone: 95%.) According to HPLC analysis, a formation ratio of cis/trans-isomers of the formed product was 5/95.

Example 6: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone

[0118] In 3 ml of methylene chloride were dissolved 0.281 g (1.0 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone and 0.072 g (1.0 mmol) of isobutyl aldehyde, and after cooling the solution to -78°C under argon atmosphere, 14 mg (0.1 mmol) of BF₃ · Et₂O was added to the solution and the mixture was reacted under stirring at -78°C for 2 hours.

[0119] Moreover, the temperature of the mixture was raised to around 0°C, and 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 10 ml of methylene chloride. The organic layer was washed with a saturated salt solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 3-diphenylmethyl-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone as colorless transparent oily substance (0.35 g). According to the HPLC analysis, a formation ratio of cis/trans isomers of the formed product was 15/85. Yield based on 3-diphenylmethyl-5-methoxy-2(3H)oxazolone was substantially quantitative.

cis-trans mixture

IR (neat, cm⁻¹): 1762 (s), 1400 (m).

¹H-NMR (δ, CDCl₃) (trans-isomer): 0.90 (d, J=6.8 Hz, 3H), 0.96 (d, J=6.4Hz, 3H), 1.90 (m, 1H), 3.38 (s, 3H), 3.96 (d, J=4.9Hz, 1H), 4.14 (dd, J=6.8Hz, 4.9Hz, 1H), 6.19 (s, 1H), 7.2-7.4 (m, 10H).

MS (EI) m/z: 353 (MH⁺), 167.

Example 7: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone

[0120] In 3 ml of methylene chloride were dissolved 0.281 g (1.0 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone and 0.134 g (1.0 mmol) of hydrocinnamaldehyde, and after cooling the solution to -78°C under argon atmosphere, 14 mg (0.1 mmol) of BF₃ · Et₂O was added to the solution and the mixture was reacted under stirring at -78°C

for 2 hours.

[0121] Moreover, the temperature of the mixture was raised to around -20°C, and 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 10 ml of methylene chloride. The organic layer was washed with a saturated salt solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. By adding n-hexane to the residue, the material was crystallized to obtain 3-diphenylmethyl-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone as white crystal (0.41 g). According to the HPLC analysis, a formation ratio of cis/trans isomers of the formed product was 5/95. Yield based on 3-diphenylmethyl-5-methoxy-2(3H)oxazolone was 99%.

cis-trans mixture

Melting point: 104 to 106 °C

IR (KBr, cm⁻¹): 1762 (s), 1399 (m).

¹H-NMR (δ, CDCl₃) (trans-isomer): 2.05 (m, 2H), 2.75 (m, 2H), 3.35 (s, 3H), 3.93 (d, J=4.4Hz, 1H), 4.36 (m, 1H), 6.21 (s, 1H), 7.15-7.37 (m, 15H).

MS (EI) m/z: 415 (M⁺), 167.

Example 8: Synthesis of 3-(1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0122] In 4 ml of methylene chloride were dissolved 0.22 g (1.0 mmol) of 3-(1-phenylethyl)-5-methoxy-2(3H)oxazolone and 0.106 g (1.0 mmol) of benzaldehyde, and after cooling the solution to -78°C under argon atmosphere, 14 mg (0.1 mmol) of BF₃ · Et₂O was added to the solution and the mixture was reacted under stirring at -78°C for 2 hours.

[0123] Moreover, the temperature of the mixture was raised to around 0°C, and 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 10 ml of methylene chloride. The organic layer was washed with a saturated salt solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 3-(1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone as colorless transparent oily substance (0.32 g). By adding n-hexane to the residue, the material was obtained as a white crystal. According to ¹H-NMR, a formation ratio of cis/trans isomers of the formed product was 1/99 and a diastereomer ratio of the trans-isomer was 70/30. Yield based on 3-(1-phenylethyl)-5-methoxy-2(3H)oxazolone was 98%.

trans isomer (diastereomer mixture)

Melting point: 98 to 106 °C

IR (KBr, cm⁻¹): 1737 (s), 1399 (m).

¹H-NMR (δ, CDCl₃) (trans-isomer, major isomer): 1.58 (d, J=6.8Hz, 3H), 3.82 (s, 3H), 3.84 (d, J=4.4Hz, 1H), 5.32 (d, J=4.4Hz, 1H), 5.30 (q, J=6.8Hz, 1H), 7.07-7.40 (m, 10H), (trans-isomer, minor isomer): 1.62 (d, J=6.8Hz, 3H), 3.32 (s, 3H), 4.13 (d, J=4.4Hz, 1H), 5.20 (q, J=6.8Hz, 1H), 5.36 (d, J=4.4Hz, 1H), 7.07-7.40 (m, 10H).

MS (CI, i-C₄H₁₀) m/z: 326 (MH⁺), 105.

Example 9: Synthesis of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0124] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 110 mg (0.5 mmol) of 3-((R)-1-phenylethyl)-5-methoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour.

[0125] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 95%. A formation ratio of cis/trans isomers of the formed product was 1/99 according to the ¹H-NMR analysis and a diastereomer ratio of the trans-isomer was 72/28.

Example 10: Synthesis of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0126] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 110 mg (0.5 mmol) of 3-((R)-1-phenylethyl)-5-methoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-t-butyltrimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour.

[0127] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrate was quantitated by the HPLC method, yield of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 85%. A formation ratio of cis/trans isomers of the formed product was 5/95 according to the ¹H-NMR analysis and a diastereomer ratio of the trans-isomer was 70/30.

Example 11: Synthesis of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0128] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 110 mg (0.5 mmol) of 3-((R)-1-phenylethyl)-5-methoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.1 ml (0.1 mmol) of a methylene chloride solution containing 1.0 M-SnCl₄ were added to the solution and the mixture was reacted under stirring for one hour.

[0129] Moreover, after the temperature of the mixture was raised up to the room temperature and stirring for further two hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrate was quantitated by the HPLC method, yield of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 78%. A formation ratio of cis/trans isomers of the formed product was 4/96 and a diastereomer ratio of the trans-isomer was 83/17.

Example 12: Synthesis of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0130] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 110 mg (0.5 mmol) of ((R)-1-phenylethyl)-5-methoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-BBr₃ were added to the solution and the mixture was reacted under stirring for one hour.

[0131] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrate was quantitated by the HPLC method, yield of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 86%. A formation ratio of cis/trans isomers of the formed product was 3/97 and a diastereomer ratio of the trans-isomer was 1/1.

Example 13: Synthesis of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0132] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 110 mg (0.5 mmol) of 3-((R)-1-phenylethyl)-5-methoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 15 mg (0.05 mmol) of trifluoromethanesulfonic acid were added to the solution and the mixture was reacted under stirring for one hour.

[0133] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrate was quantitated by the HPLC method, yield of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 67%. A formation ratio of cis/trans isomers of the formed product was 4/96 and a diastereomer ratio of the trans-isomer was 67/33.

Example 14: Synthesis of 3-((S)-1-phenylethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone

[0134] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 124 mg (0.5 mmol) of 3-((S)-1-phenylethyl)-5-isopropoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour.

[0135] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of

methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 3-((S)-1-phenylethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone (171mg (0.485 mmol)) as a brownish oily substance. A formation ratio of cis/trans isomers of the formed product was 1/99 according to ¹H-NMR and a diastereomer ratio of the trans-isomer was 75/25. Yield based on 3-((S)-1-phenylethyl)-5-isopropoxy-2(3H)oxazolone was 97%.

trans isomer (diastereomer mixture)

MS (CI, i-C₄H₁₀) m/z: 354 (MH⁺).

¹H-NMR (δ, CDCl₃) (trans-isomer, major isomer): 1.30 (d, J=6.3Hz, 6H), 1.60 (d, J=7.3Hz, 3H), 3.77 (d, J=4.4Hz, 1H), 5.14 (m, 1H), 5.28 (d, J=4.4Hz, 1H), 5.30 (q, J=7.3Hz, 1H), 7.24-7.39 (m, 10H), (trans-isomer, minor isomer): 1.08 (d, J=6.4Hz, 6H), 1.65 (d, J=6.8Hz, 3H), 4.08 (d, J=4.4Hz, 1H), 4.70 (m, 1H), 5.06 (q, J=6.8Hz, 1H), 5.31 (d, J=4.4Hz, 1H), 7.24-7.39 (m, 10H).

Example 15: Synthesis of 3-((S)-1-phenylethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone

[0136] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 124 mg (0.5 mmol) of 3-((S)-1-phenylethyl)-5-isopropoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-BF₃·Et₂O were added to the solution and the mixture was reacted under stirring for one hour.

[0137] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrated residue was quantitated by the HPLC method, yield of 3-((S)-1-phenylethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone was 82%. According to ¹H-NMR, a formation ratio of cis/trans isomers of the formed product was 4/96 and a diastereomer ratio of the trans-isomer was 69/31.

Example 16: Synthesis of 3-((S)-1-phenylethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone

[0138] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 124 mg (0.5 mmol) of 3-((S)-1-phenylethyl)-5-isopropoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-TiCl₄ were added to the solution and the mixture was reacted under stirring for 2 hours.

[0139] Moreover, after the temperature of the mixture was raised up to the room temperature and stirring for further one hour, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrated residue was quantitated by the HPLC method, yield of 3-((S)-1-phenylethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone was 59%. According to ¹H-NMR analysis, no formation of a cis isomer was admitted and a diastereomer ratio of the trans-isomer was 50/50.

Example 17: Synthesis of 3-((S)-1-phenylethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone

[0140] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 124 mg (0.5 mmol) of ((S)-1-phenylethyl)-5-isopropoxy-2(3H)oxazolone, and after cooling the solution to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-AlCl₃ were added to the solution and the mixture was reacted under stirring for one hour.

[0141] After the temperature of the mixture was raised up to the room temperature and stirring for further one hour, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the resulting concentrated residue was quantitated by the HPLC method, yield of 3-((S)-1-phenylethyl)-4-isopropoxycarbonyl-5-phenyl-2-oxazolidinone was 68%. According to ¹H-NMR analysis, a formation ratio of cis/trans isomers of the formed product was 5/95 and a diastereomer ratio of the trans-isomer was 66/34.

Example 18: Synthesis of 3-(1-phenylethyl)-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone

[0142] In 6 ml of methylene chloride were dissolved 0.438 g (2.0 mmol) of 3-(1-phenylethyl)-5-methoxy-2(3H)oxa-

zalone and 0.151 g (2.1 mmol) of isobutyl aldehyde, and the solution was cooled to -78°C under argon atmosphere, 14 mg (0.1 mmol) of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ was added to the solution and the mixture was reacted under stirring at -78°C for 30 minutes. Thereafter, the temperature was raised up to -40°C and the mixture was reacted under stirring for one hour.

[0143] Moreover, the temperature of the mixture was raised up to around -20°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 3-(1-phenylethyl)-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone as a yellowish transparent oily substance (0.57 g). According to ^1H -NMR analysis, a formation ratio of cis/trans isomers of the formed product was 7/93 and a diastereomer ratio of the trans-isomer was 80/20. Yield based on 3-(1-phenylethyl)-5-methoxy-2(3H)oxazolone was 96%.

trans isomer (diastereomer mixture)

IR (neat, cm^{-1}): 1760 (s), 1406 (m).

^1H -NMR (δ , CDCl_3) (trans-isomer, major isomer): 0.73 (d, $J=6.8\text{Hz}$, 3H), 0.76 (d, $J=6.8\text{Hz}$, 3H), 1.55 (d, $J=7.3\text{Hz}$, 3H), 1.67 (m, 1H), 3.60 (d, $J=4.4\text{Hz}$, 1H), 3.78 (s, 3H), 4.80 (dd, $J=4.48\text{Hz}$, $J=5.9\text{Hz}$, 1H), 5.27 (q, $J=7.3\text{Hz}$, 1H), 7.2-7.4 (m, 5H), (trans-isomer, minor isomer): 0.94 (d, $J=6.8\text{Hz}$, 3H), 0.97 (d, $J=6.8\text{Hz}$, 3H), 1.67 (d, $J=7.3\text{Hz}$, 3H), 1.88 (m, 1H), 3.24 (s, 3H), 3.98 (d, $J=4.4\text{Hz}$, 1H), 4.11 (dd, $J=4.4\text{Hz}$, $J=5.9\text{Hz}$, 1H), 5.14 (q, $J=7.3\text{Hz}$, 1H).

MS (EI) m/z : 291 (M^+), 105.

Example 19: Synthesis of 3-(1-phenylethyl)-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone

[0144] In 4 ml of methylene chloride were dissolved 0.22 g (1.0 mmol) of 3-(1-phenylethyl)-5-methoxy-2(3H)oxazolone and 0.134 g (1.0 mmol) of hydrocinnamaldehyde, and the solution was cooled to -78°C under argon atmosphere, 14mg (0.1 mmol) of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ was added to the solution and the mixture was reacted under stirring at -78°C for 30 minutes. Thereafter, the temperature was raised up to -40°C and the mixture was reacted under stirring for one hour.

[0145] Moreover, the temperature of the mixture was raised up to around -20°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure, and the residue was applied to silica gel column chromatography (eluent: n-hexane/ethyl acetate = 5/1) to obtain 3-(1-phenylethyl)-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone as a yellowish transparent oily substance (0.30 g). A formation ratio of cis/trans isomers of the formed product was 7/93, and according to ^1H -NMR analysis, a diastereomer ratio of the trans-isomer was 86/14. Yield based on 3-(1-phenylethyl)-5-methoxy-2(3H)oxazolone was 84%.

trans isomer (diastereomer mixture)

IR (neat, cm^{-1}): 1760 (s), 1404 (m).

^1H -NMR (δ , CDCl_3) (trans-isomer, major isomer): 1.56 (d, $J=7.3\text{Hz}$, 3H), 1.76 (m, 2H), 2.58 (t, $J=7.8\text{Hz}$, 2H), 3.55 (d, $J=3.9\text{Hz}$, 1H), 4.30 (m, 1H), 5.28 (q, $J=7.8\text{Hz}$, 1H), 7.1-7.4 (m, 10H).

MS (EI) m/z : 353 (M^+), 105.

Example 20: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0146] In 4 ml of methylene chloride were dissolved 0.27 g (1.0 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 0.106g (1.0 mmol) of benzaldehyde, and the solution was cooled to -78°C under argon atmosphere, 14 mg (0.1 mmol) of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ was added to the solution and the mixture was reacted under stirring for 2 hours.

[0147] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone as a colorless transparent oily substance (0.37 g). According to ^1H -NMR analysis, no formation of a cis isomer was admitted and a diastereomer ratio of the trans-isomer was 72/28. Yield based on 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 99%.

trans isomer (diastereomer mixture)

IR (neat, cm^{-1}): 1752 (s), 1402 (m).

^1H -NMR (δ , CDCl_3) (trans-isomer, major isomer): 1.83 (d, $J=6.8\text{Hz}$, 3H), 3.30 (d, $J=4.9\text{Hz}$, 1H), 3.73 (s, 3H), 5.19 (d, $J=4.9\text{Hz}$, 1H), 6.00 (q, $J=6.8\text{Hz}$, 1H), 6.8-7.9 (m, 12H), (trans-isomer, minor isomer): 1.67 (d, $J=6.8\text{Hz}$, 3H), 2.61 (s, 3H), 4.11 (d, $J=4.4\text{Hz}$, 1H), 5.20 (d, $J=4.4\text{Hz}$, 1H), 6.09 (q, $J=6.8\text{Hz}$, 1H).

MS (EI) m/z: 375 (M⁺), 155, 106.

Example 21: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0148] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 135 mg (0.5 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour.

[0149] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 92%. A formation ratio of cis/trans isomers of the formed product was 1/99, and a diastereomer ratio of the trans-isomer was 86/14.

Example 22: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0150] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 135 mg (0.5 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.1 ml (0.1 mmol) of a methylene chloride solution containing 1.0 M-SnCl₄ were added to the solution and the mixture was raised to the room temperature and reacted under stirring for 20 hours.

[0151] To the resulting reaction mixture was added 15 ml of a saturated aqueous sodium hydrogen carbonate solution and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone was 54%. A formation ratio of cis/trans isomers of the formed product was 5/95, and a diastereomer ratio of the trans-isomer was 81/19.

Example 23: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone

[0152] In 4 ml of methylene chloride were dissolved 0.27 g (1.0 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 0.134 g (1.0 mmol) of hydrocinnamaldehyde, and under argon atmosphere, the solution was cooled to -78°C, and 14 mg (0.1 mmol) of BF₃ · Et₂O was added to the solution and the mixture was reacted under stirring at -78°C for 2 hours.

[0153] Moreover, the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. By adding n-hexane to the residue, the material was crystallized to obtain 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(2-phenylethyl)-2-oxazolidinone as white crystal (0.36 g). According to ¹H-NMR analysis, a formation ratio of the cis/trans isomers of the formed product was 1/99 and a diastereomer ratio of the trans-isomer was 97/3. Yield based on 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 99%.

trans isomer (diastereomer mixture)

Melting point: 145 to 147 °C

IR (neat, cm⁻¹): 1745 (s), 1414 (m).

¹H-NMR (δ, CDCl₃) (trans-isomer, major isomer): 1.45 (m, 2H), 1.79 (d, J=6.8Hz, 3H), 2.38 (t, J=7.8Hz, 2H), 2.99 (d, J=3.9Hz, 1H), 3.69 (s, 3H), 4.19 (m, 1H), 5.94 (q, J=6.8Hz, 1H), 6.8-7.9 (m, 12H).

MS (EI) m/z: 403 (M⁺), 155.

Example 24: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone

[0154] In 3 ml of methylene chloride were dissolved 0.141 g (0.52 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 0.04 g (0.55 mmol) of isobutyl aldehyde, and under argon atmosphere, the solution was cooled to -78°C, and 10 mg of BF₃ · Et₂O was added to the solution and the mixture was reacted under stirring at -78°C for 2 hours.

[0155] Moreover, the temperature of the mixture was raised up to -20°C, 15 ml of a saturated aqueous sodium

hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. By adding n-hexane to the residue, the material was crystallized to obtain 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-isopropyl-2-oxazolidinone as white crystal (0.17 g). According to $^1\text{H-NMR}$, a formation ratio of the cis/trans isomers of the formed product was 1/99 and a diastereomer ratio of the trans-isomer was 93/7. Moreover, by washing the crystal with n-hexane, 0.12 g of a main isomer of the trans-isomer could be obtained with a purity of 99%. Yield based on 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 96%.

trans isomer (major isomer)

Melting point: 102 to 106 °C

IR (neat, cm^{-1}): 1751 (s), 1732 (s), 1413 (m).

$^1\text{H-NMR}$ (δ , CDCl_3): 1.45 (m, 2H), 1.79 (d, $J=6.8\text{Hz}$, 3H), 2.38 (t, $J=7.8\text{Hz}$, 2H), 2.99 (d, $J=3.9\text{Hz}$, 1H), 3.69 (s, 3H), 4.19 (m, 1H), 5.94 (q, $J=6.8\text{Hz}$, 1H), 6.8-7.9 (m, 12H).

MS (EI) m/z : 341 (M^+), 155.

Example 25: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-tridecyl-2-oxazolidinone

[0156] In 4 ml of methylene chloride were dissolved 0.438 g (2.0 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone and 0.424 g (2.0 mmol) of tetradecyl aldehyde, and under argon atmosphere, the solution was cooled to -20°C, and 28mg (0.2 mmol) of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ was added to the solution and the mixture was reacted under stirring at the state for 2 hours and further at 0°C for one hour.

[0157] To the resulting reaction mixture was added 15 ml of a saturated aqueous sodium hydrogen carbonate solution and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. By adding 5 ml of methanol to the obtained residue, the material was crystallized to obtain 0.6 g of 3-diphenylmethyl-4-methoxycarbonyl-5-tridecyl-2-oxazolidinone as white crystal. According to HPLC analysis, a formation ratio of the cis/trans isomers of the formed product was 10/90. Yield based on 3-diphenylmethyl-5-methoxy-2(3H)oxazolone was 70%.

cis-trans isomers (mixture)

Melting point: 61 to 65 °C

IR (KBr, cm^{-1}): 1772 (s), 1740 (s), 1393 (m).

$^1\text{H-NMR}$ (δ , CDCl_3) (trans isomer): 0.88 (t, $J=6.6\text{Hz}$, 3H), 1.1-1.7 (m, 24H), 3.37 (s, 3H), 3.92 (d, $J=4.4\text{Hz}$, 4.33 (m, 1H), 6.20 (s, 1H), 7.2-7.4 (m, 10H).

MS (EI) m/z : 493 (M^+), 167.

Example 26: Synthesis of 3-diphenylmethyl-4-((*l*)-menthyloxy)carbonyl-5-phenyl-2-oxazolidinone

[0158] In 4 ml of methylene chloride were dissolved 0.405 g (1.0 mmol) of 3-diphenylmethyl-5-((*l*)-menthyloxy)-2(3H)oxazolone and 0.106 g (1.0 mmol) of benzaldehyde, and under argon atmosphere, the solution was cooled to -78°C, and TiCl_4 (19 mg; 0.1 mmol) was added to the solution and the mixture was reacted under stirring at the state for one hour and further at -20°C for 2 hours.

[0159] Moreover, the temperature of the mixture was raised to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure, and the residue was applied to silica gel column chromatography (eluent: n-hexane/ethyl acetate = 9/1) to obtain 0.24 g of a diastereomer mixture of a trans-isomer of 3-diphenylmethyl-4-((*l*)-menthyloxy)carbonyl-5-phenyl-2-oxazolidinone as white crystal. From $^1\text{H-NMR}$, a diastereomer ratio of the trans isomer was 64/36. Yield based on 3-diphenylmethyl-5-((*l*)-menthyloxy)-2(3H)oxazolone was 47%.

trans isomer (diastereomer mixture)

Melting point: 136 to 139 °C

IR (KBr, cm^{-1}): 1757 (s), 1390 (m).

$^1\text{H-NMR}$ (δ , CDCl_3) (major isomer): 4.18 (d, $J=3.7\text{Hz}$, 1H), 5.36 (d, $J=3.7\text{Hz}$, 1H), 6.12 (s, 1H), (minor isomer): 4.15 (d, $J=3.7\text{Hz}$, 1H), 5.39 (d, $J=3.7\text{Hz}$, 1H), 6.18 (s, 1H).

MS (CI, $i\text{-C}_4\text{H}_{10}$) m/z : 512 (MH^+).

Example 27: Synthesis of 3-diphenylmethyl-4-((*l*)-menthyloxy)carbonyl-5-phenyl-2-oxazolidinone

[0160] In 4 ml of methylene chloride were dissolved 0.405 g (1.0 mmol) of 3-diphenylmethyl-5-((*l*)-menthyloxy)-2(3H)oxazolone and 0.106 g (1.0 mmol) of benzaldehyde, and under argon atmosphere, the solution was cooled to -78°C, and BF₃ · Et₂O (15 mg; 0.1 mmol) was added to the solution and the mixture was reacted under stirring at the state for 4 hours.

[0161] Moreover, the temperature of the mixture was raised to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. By adding 15 ml of methanol to the obtained residue, the material was crystallized to obtain 0.24 g of a diastereomer mixture of a trans-isomer of 3-diphenylmethyl-4-((*l*)-menthyloxy)carbonyl-5-phenyl-2-oxazolidinone as white crystal. The selectivity was reversed to the case of Example 11, a diastereomer ratio of the formed product was 48/52. Yield based on 3-diphenylmethyl-5-((*l*)-menthyloxy)-2(3H)oxazolone was 47%.

Example 28: Synthesis of 3-diphenylmethyl-4-((1*S*,2*S*,5*R*)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-phenyl-2-oxazolidinone

[0162] In 4 ml of methylene chloride were dissolved 3-diphenylmethyl-5-((1*S*,2*S*,5*R*)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone (0.24 g; 0.5 mmol) and benzaldehyde (0.053g; 0.5 mmol) of, and under argon atmosphere, the solution was cooled to -78°C, and 10 mg of BF₃ · Et₂O was added to the solution and the mixture was reacted under stirring at -78°C for 2 hours.

[0163] Moreover, the temperature of the mixture was raised to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. The material was crystallized to obtain 0.29 g of 3-diphenylmethyl-4-((1*S*,2*S*,5*R*)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-phenyl-2-oxazolidinone as yellowish crystal. According to ¹H-NMR analysis, a formation ratio of the cis/trans isomers of the formed product was 64/36. Also, a cis isomer and a trans isomer were obtained each as a single diastereomer, respectively. Yield based on 3-diphenylmethyl-5-((1*S*,2*S*,5*R*)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone was 99%.

cis isomer

¹H-NMR (δ, CDCl₃): 3.70 (d, J=9.8Hz, 1H), 4.13 (td, J=3.90Hz, J=10.7Hz, 1H), 5.24 (d, J=9.8Hz, 1H), 5.91 (s, 1H).

trans isomer

¹H-NMR (δ, CDCl₃): 3.50 (d, J=3.9Hz, 1H), 4.73 (td, J=3.9Hz, J=10.7Hz, 1H), 5.07 (d, J=3.9Hz, 1H), 5.97 (s, 1H).

MS (CI, i-C₄H₁₀) m/z: 588 (M⁺).

Example 29: Synthesis of 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-styryl-2-oxazolidinone

[0164] In 4 ml of methylene chloride were dissolved 0.135 g (0.5 mmol) of 3-(1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 0.066 g (0.5 mmol) of cinnamaldehyde, and after cooling the solution to -78°C under argon atmosphere, 10 mg of trimethylsilyl triflate was added to the solution and the mixture was reacted under stirring at that state for 2 hours.

[0165] Moreover, the temperature of the mixture was raised to the room temperature, and 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 10 ml of methylene chloride. The organic layer was washed with a saturated salt solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 3-(1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-styryl-2-oxazolidinone (0.20 g) as yellowish crystal. According to ¹H-NMR, a formation ratio of cis/trans isomers of the formed product was 3/97 and a diastereomer ratio of the trans isomer was 92/8. Yield based on 3-(1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 99%.

trans isomer (diastereomer mixture)

Melting point: 102 to 110 °C

IR (KBr, cm⁻¹): 1761 (s), 1683 (s), 1386 (s).

¹H-NMR (δ, CDCl₃) (major isomer): 1.81 (d, J=6.9Hz, 3H), 3.14 (d, J=3.9Hz, 1H), 3.74 (s, 3H), 4.81 (dd, J=3.9Hz, J=6.8 Hz, 1H), 5.47 (dd, J=6.8 Hz, J=15.6Hz, 1H), 5.96 (q, J=6.9Hz, 1H), 6.40 (d, J=15.6Hz, 1H), 6.9-8.0 (m, 12H).

MS (EI) m/z: 401 (M⁺), 155.

Example 30: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone

[0166] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 148 mg (0.5 mmol) of 3-diphenylmethyl-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour.

[0167] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 191 mg (0.475 mmol) of 3-diphenylmethyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone as brownish oily substance. Yield based on 3-diphenylmethyl-4-methyl-5-methoxy-2(3H)oxazolone was 95%. According to the HPLC analysis, a formation ratio of the cis/trans isomers of the product was 10/90.

cis-trans isomers (mixture)

MS (CI, i-C₄H₁₀) m/z: 402 (MH⁺).

trans isomer

¹H-NMR (δ, CDCl₃): 0.91 (s, 3H), 3.44 (s, 3H), 5.52 (s, 1H), 5.65 (s, 1H), 7.14-7.33 (m, 15H).

¹³C-NMR (δ, CDCl₃): 19.3, 52.9, 61.5, 68.8, 80.8, 126.3, 127.5, 127.7, 128.5, 128.6, 128.7, 129.0, 134.6, 138.6, 139.3, 155.9, 172.1.

cis isomer

¹H-NMR (δ, CDCl₃): 1.51 (s, 3H), 2.84 (s, 3H), 5.20 (s, 1H), 5.68 (s, 1H), 7.14-7.33 (m, 15H).

¹³C-NMR (δ, CDCl₃): 19.3, 52.9, 61.5, 68.8, 80.8, 126.3, 127.5, 127.7, 128.5, 128.6, 128.7, 129.0, 134.6, 138.6, 139.3, 155.9, 172.1.

Example 31: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone

[0168] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 148 mg (0.5 mmol) of 3-diphenylmethyl-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-t-butyldimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour.

[0169] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 3-diphenylmethyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone was 90%. According to the HPLC analysis, a formation ratio of cis/trans isomers of the formed product was 11/89.

Example 32: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone

[0170] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 148 mg (0.5 mmol) of 3-diphenylmethyl-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-TiCl₄ were added to the solution and the mixture was reacted under stirring for one hour.

[0171] Moreover, after the temperature of the mixture was raised up to the room temperature and stirring was carried out for further 18 hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 3-diphenylmethyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone was 59%. According to the HPLC analysis, a formation ratio of cis/trans isomers of the formed product was 20/80.

Example 33: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone

[0172] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 141 mg (0.5 mmol) of 3-diphenylmethyl-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-BF₃ · Et₂O were added to the solution and the mixture was reacted under stirring for one hour.

[0173] Moreover, after the temperature of the mixture was raised up to the room temperature and stirring was carried out for further 15 hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 3-diphenylmethyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone was 43%. According to the HPLC analysis, a formation ratio of cis/trans isomers of the formed product was 31/69.

Example 34: Synthesis of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone

[0174] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 117 mg (0.5 mmol) of 3-((R)-1-phenylethyl)-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour.

[0175] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 154 mg (0.455 mmol) of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone as brownish oily substance. A diastereomer ratio of the formed product was (1):(2):(3):(4)=46:30:20:4. Yield based on 3-((R)-1-phenylethyl)-4-methyl-5-methoxy-2(3H)oxazolone was 91%.

diastereomer mixture

MS (CI, i-C₄H₁₀) m/z: 340 (MH⁺).

Formed product (1) (trans isomer)

¹H-NMR (δ, CDCl₃): 0.92 (s, 3H), 1.78 (d, J=7.3Hz, 3H), 3.79 (s, 3H), 4.55 (q, J=7.3Hz, 1H), 5.57 (s, 1H), 7.18-7.45 (m, 10H).

Formed product (2) (trans isomer)

¹H-NMR (δ, CDCl₃): 0.95 (s, 3H), 1.89 (d, J=7.3Hz, 3H), 3.57 (s, 3H), 4.67 (q, J=7.3Hz, 1H), 5.62 (s, 1H), 7.18-7.45 (m, 10H).

Formed product (3) (cis isomer)

¹H-NMR (δ, CDCl₃): 1.48 (s, 3H), 1.70 (d, J=7.8Hz, 3H), 3.10 (s, 3H), 5.01 (q, J=7.8Hz, 1H), 5.15 (s, 1H), 7.18-7.45 (m, 10H).

Formed product (4) (cis isomer)

¹H-NMR (δ, CDCl₃): 1.54 (d, J=7.3Hz, 3H), 1.60 (s, 3H), 3.17 (s, 3H), 4.90 (q, J=7.3Hz, 1H), 5.17 (s, 1H), 7.18-7.45 (m, 10H).

Example 35: Synthesis of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone

[0176] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 117 mg (0.5 mmol) of 3-((R)-1-phenylethyl)-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.1 ml (0.1 mmol) of methylene chloride solution containing 1.0 M-SnCl₄ were added to the solution and the mixture was reacted under stirring for one hour.

[0177] Moreover, after the temperature of the mixture was raised up to the room temperature and stirring was carried out further for 3 hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 74.7 mg (0.22 mmol) of 3-((R)-1-phenylethyl)-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone as brownish oily substance. A diastereomer ratio of the formed product was (1):(2):(3):(4)=42:13:39:6. Yield based on 3-((R)-1-phenylethyl)-4-methyl-5-methoxy-2(3H)oxazolone was 44%.

Example 36: Synthesis of 3-benzyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone

[0178] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 110 mg (0.5 mmol) of 3-benzyl-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour.

[0179] Moreover, after the temperature of the mixture was raised up to 0°C, 15 ml of a saturated aqueous sodium

hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 158 mg (0.485 mmol) of 3-benzyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone as brownish oily substance. Yield based on 3-benzyl-4-methyl-5-methoxy-2(3H)oxazolone was 97%. A formation ratio of the cis/trans isomers of the formed product was 59/41.

cis-trans isomers (mixture)

MS (CI, $i\text{-C}_4\text{H}_{10}$) m/z : 326 (MH^+).

$^1\text{H-NMR}$ (δ , CDCl_3) (trans isomer): 0.92 (s, 3H), 3.59 (s, 3H), 4.46 (d, $J=15.6\text{Hz}$, 1H), 4.57 (d, $J=15.6\text{Hz}$, 1H), 5.66 (s, 1H), 7.24-7.37 (m, 10H), (cis isomer): 1.53 (s, 3H), 3.14 (s, 3H), 4.08 (d, $J=15.6\text{Hz}$, 1H), 4.77 (d, $J=15.6\text{Hz}$, 1H), 5.21 (s, 1H), 7.24-7.37 (m, 10H).

Example 37: Synthesis of 3-benzyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone

[0180] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 110 mg (0.5 mmol) of 3-benzyl-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C , 53.1 mg (0.5 mmol) of benzaldehyde and 0.1 ml (0.1 mmol) of a methylene chloride solution containing 1.0 M- SnCl_4 were added to the solution and the mixture was reacted under stirring for one hour.

[0181] Moreover, after the temperature of the mixture was raised up to the room temperature and stirring was carried out for 3 hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 3-benzyl-4-methoxycarbonyl-4-methyl-5-phenyl-2-oxazolidinone was 62%. A formation ratio of the cis/trans isomers of the formed product was 34/66.

Example 38: Synthesis of 4,5-anti-3-diphenylmethyl-4,5-dimethoxycarbonyl-2-oxazolidinone

[0182] Under argon atmosphere, in 2 ml of methylene chloride was dissolved 141 mg (0.5 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C , 1 ml of a methylene chloride solution containing 0.5M methyl glyoxylate (44 mg (0.5 mmol) as methyl glyoxylate) and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for one hour. Moreover, the temperature of the mixture was raised up to the room temperature and stirring was carried out for 15 hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (n-hexane:ethyl acetate:methanol = 5:1:1) to obtain 51.7 mg (0.14 mmol) of 4,5-anti-3-diphenylmethyl-4,5-dimethoxycarbonyl-2-oxazolidinone as colorless transparent oily substance. Yield based on 3-diphenylmethyl-5-methoxy-2(3H)-oxazolone was 28%.

MS (CI, $i\text{-C}_4\text{H}_{10}$) m/z : 370 (MH^+).

$^1\text{H-NMR}$ (δ , CDCl_3): 3.38 (s, 3H), 3.86 (s, 3H), 4.40 (d, $J=2.9\text{Hz}$, 1H), 4.83 (d, $J=2.9\text{Hz}$, 1H), 6.24 (s, 1H), 5.65 (s, 1H), 7.20-7.36 (m, 10H).

Example 39: Synthesis of 4,5-anti-3-diphenylmethyl-4,5-dimethoxycarbonyl-2-oxazolidinone

[0183] Under argon atmosphere, in 2 ml of methylene chloride was dissolved 141 mg (0.5 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C , 1 ml of a methylene chloride solution containing 0.5M methyl glyoxylate (44mg (0.5mmol)) and 0.05ml (0.05 mmol) of a methylene chloride solution containing 1.0 M- $\text{BF}_3 \cdot \text{Et}_2\text{O}$ were added to the solution and the mixture was reacted under stirring for 2 hours.

[0184] Moreover, the temperature of the mixture was raised up to the room temperature and stirring was carried out for 16 hours, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 4,5-anti-3-diphenylmethyl-4,5-dimethoxycarbonyl-2-oxazolidinone was 47%.

Example 40: Synthesis of 4,5-anti-3-diphenylmethyl-4,5-dimethoxycarbonyl-2-oxazolidinone

[0185] Under argon atmosphere, in 2 ml of methylene chloride was dissolved 141 mg (0.5 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 1 ml of a methylene chloride solution containing 0.5M methyl glyoxylate (44mg(0.5 mmol)) and 0.05 ml (0.05 mmol) of a methylene chloride solution containing 1.0 M-TiCl₄ were added to the solution and the mixture was reacted under stirring for one hour.

[0186] Moreover, the temperature of the mixture was raised up to the room temperature and stirring was carried out for one hour, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with 15 ml of a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. When the obtained concentrated residue was quantitated by the HPLC method, yield of 4,5-anti-3-diphenylmethyl-4,5-dimethoxycarbonyl-2-oxazolidinone was 48%.

Example 41: Synthesis of 3-diphenylmethyl-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone

[0187] In 3 ml of methylene chloride was dissolved 141 mg (0.5 mmol) of 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, and 75 mg (0.5 mmol) of 3,4-methylenedioxybenzaldehyde was added to the solution. After the solution was cooled to -78°C under argon atmosphere, 5 ml (0.05 mmol) of a 0.1 N-methylene chloride solution containing BF₃ · Et₂O was added to the mixture and the mixture was reacted at that state under stirring for 30 minutes. Moreover, the temperature of the mixture was raised to 0°C and the mixture was reacted under stirring for one hour.

[0188] Furthermore, the temperature of the mixture was raised up to around the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 180 mg of 3-diphenylmethyl-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone as white crystal. According to the ¹H-NMR analysis, no formation of a cis isomer was admitted. Yield based on 3-diphenylmethyl-5-methoxy-2(3H)oxazolone was substantially quantitative.

¹H-NMR (δ, CDCl₃) (trans isomer): 3.40 (s, 3H), 4.16 (d, J=3.9Hz, 1H), 5.34 (d, J=3.9Hz, 1H), 6.00 (s, 2H), 6.26 (s, 1H), 6.78 (m, 3H), 7.23-7.33 (m, 10H).

MS (EI) m/z: 431 (M⁺), 206, 167.

Elemental analysis:			
Calcd:	C, 69.60;	H, 4.91;	N, 3.25
Found:	C, 69.50;	H, 4.93;	N, 3.20.

Example 42: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-dimethoxyphenyl)-2-oxazolidinone

[0189] In 3 ml of methylene chloride were dissolved 0.135 g (0.5 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 0.083 g (0.5 mmol) of 3,4-dimethoxybenzaldehyde, and after cooling the solution to -78°C under argon atmosphere, 0.5 ml (0.05 mmol) of 0.1 N-methylene chloride solution of trimethylsilyl triflate was added to the solution and the mixture was reacted at that state under stirring for 30 minutes.

[0190] Moreover, the temperature of the mixture was raised up to around the room temperature, and 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure, and applied to silica gel column chromatography (eluent: n-hexane/ethyl acetate = 4/1) to obtain 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-phenyl-2-oxazolidinone as white crystal (0.19 g). A formation ratio of cis/trans isomers of the formed product was 2/98, and according to the ¹H-NMR analysis, a diastereomer ratio of the trans isomer was 86/14. Yield based on 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 86%.

trans isomer (diastereomer mixture)

IR (KBr, cm⁻¹): 1755 (s), 1518 (m), 1399 (m).

¹H-NMR (δ, CDCl₃) (major isomer): 1.81 (d, J=6.8 Hz, 3H), 3.39 (d, J=5.4Hz, 1H), 3.42 (s, 3H), 3.75 (s, 3H), 3.77

(s, 3H), 5.14 (d, J=5.4Hz, 1H), 6.04 (q, J=6.8Hz, 1H), 6.24 (d, J=2.0Hz, 1H), 6.46 (dd, J=2.0Hz, J=8.3Hz, 1H), 6.54 (d, J=8.3Hz, 1H), 7.38-7.55 (m, 4H), 7.83 (m, 2H), 8.04 (d, J=8.3Hz, 1H).

MS (EI) m/z: 435 (M⁺), 222, 155.

Elemental analysis:

Calcd:	C, 65.98;	H, 5.79;	N, 3.22
Found:	C, 68.66;	H, 5.87;	N, 3.30.

Example 43: Synthesis of 3-((R)-1-phenylethyl)-4-((S)-menthyloxycarbonyl)-5-phenyl-2-oxazolidinone

[0191] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 171.7 mg (0.5 mmol) of 3-((R)-1-phenylethyl)-5-((S)-menthyloxy)-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for 30 minutes.

[0192] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 211.5 mg (0.47 mmol) of 3-((R)-1-phenylethyl)-4-((S)-menthyloxycarbonyl)-5-phenyl-2-oxazolidinone as pale yellowish crystal. According to the ¹H-NMR analysis, a formation ratio of the cis/trans isomers of the formed product was 6/94 and a diastereomer ratio of the trans isomer was 61/39. Yield based on 3-((R)-1-phenylethyl)-5-((S)-menthyloxy)-2(3H)oxazolone was 93%.

trans isomer (diastereomer mixture)

MS (CI, i-C₄H₁₀) m/z: 450 (MH⁺).

¹H-NMR (δ, CDCl₃) (major isomer): 0.70-2.04 (m, 21H), 3.77 (d, J=4.4 Hz, 1H), 4.83 (ddd, J=4.4Hz, 1H), 5.22 (d, J=4.4Hz, 1H), 5.30 (q, J=6.8Hz, 1H), 7.09-7.39 (m, 10H), (minor isomer): 0.70-2.04 (m, 21H), 4.09 (d, J=4.4 Hz, 1H), 4.61 (ddd, J=4.4Hz, 1H), 4.93 (q, J=7.6Hz, 1H), 5.29 (d, J=4.4Hz, 1H), 7.09-7.39 (m, 10H).

Example 44: Synthesis of 3-((R)-1-phenylethyl)-4-((S)-menthyloxycarbonyl)-5-phenyl-2-oxazolidinone

[0193] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 171.7 mg (0.5 mmol) of 3-((R)-1-phenylethyl)-5-((S)-menthyloxy)-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M boron trifluoride diethyl etherate were added to the solution and the mixture was reacted under stirring for one hour.

[0194] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 209 mg (0.46 mmol) of 3-((R)-1-phenylethyl)-4-((S)-menthyloxycarbonyl)-5-phenyl-2-oxazolidinone as pale yellowish crystal. According to the ¹H-NMR analysis, a formation ratio of the cis/trans isomers of the formed product was 14/86 and a diastereomer ratio of the trans isomer was 72/28. Yield based on 3-((R)-1-phenylethyl)-5-((S)-menthyloxy)-2(3H)oxazolone was 93%.

Example 45: Synthesis of 3-((S)-1-phenylethyl)-4-((S)-menthyloxycarbonyl)-5-phenyl-2-oxazolidinone

[0195] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 171.7 mg (0.5 mmol) of 3-((S)-1-phenylethyl)-5-((S)-menthyloxy)-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for 30 minutes.

[0196] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 207 mg (0.45 mmol) of 3-((S)-1-phenylethyl)-4-((S)-menthyloxycarbonyl)-5-phenyl-2-oxazolidinone as pale yellowish crystal. According to the ¹H-NMR analysis, a diastereomer ratio of a trans isomer was 84/16. Yield based on 3-((S)-1-phenylethyl)-5-((S)-menthyloxy)-2(3H)oxa-

zolone was 92%.

trans isomer (diastereomer mixture)

MS (CI, $i\text{-C}_4\text{H}_{10}$) m/z : 450 (MH^+).

$^1\text{H-NMR}$ (δ , CDCl_3) (major isomer): 0.68-2.04 (m, 21H), 3.80 (d, $J=4.4$ Hz, 1H), 4.77 (ddd, $J=4.4$ Hz, 1H), 5.25 (d, $J=4.4$ Hz, 1H), 5.31 (q, $J=6.8$ Hz, 1H), 7.23-7.38 (m, 10H), (minor isomer): 0.68-2.04 (m, 21H), 4.1 (d, $J=4.4$ Hz, 1H), 4.58 (ddd, $J=4.4$ Hz, 1H), 4.95 (q, $J=7.6$ Hz, 1H), 5.29 (d, $J=4.4$ Hz, 1H), 7.23-7.38 (m, 10H).

Example 46: Synthesis of 3-((S)-1-phenylethyl)-4-((ℓ)-menthyloxycarbonyl)-5-phenyl-2-oxazolidinone

[0197] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 171.7 mg (0.5 mmol) of 3-((S)-1-phenylethyl)-5-((ℓ)-menthyloxy)-2(3H)oxazolone, and after the solution was cooled to -78°C , 53.1 mg (0.5 mmol) of benzaldehyde and 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M boron trifluoride diethyl etherate were added to the solution and the mixture was reacted under stirring for one hour.

[0198] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 199 mg (0.44 mmol) of 3-((S)-1-phenylethyl)-4-((ℓ)-menthyloxycarbonyl)-5-phenyl-2-oxazolidinone as pale yellowish crystal. According to the $^1\text{H-NMR}$ analysis, a formation ratio of the cis/trans isomers of the formed product was 20/80 and a diastereomer ratio of the trans isomer was 70/30. Yield based on 3-((S)-1-phenylethyl)-5-((ℓ)-menthyloxy)-2(3H)oxazolone was 89%.

Example 47: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone

[0199] Under argon atmosphere, in 3 ml of methylene chloride were dissolved 134.7 mg (0.5 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 75.1 mg (0.5 mmol) of 3,4-methylenedioxybenzaldehyde, and after the solution was cooled to -78°C , 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M-trimethylsilyl triflate was added to the solution and the mixture was reacted under stirring for 2 hours.

[0200] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure, and applied to silica gel column chromatography (eluent: n-hexane/ethyl acetate = 5.5/1) to obtain 190.4mg (0.44 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone as white crystal. According to the $^1\text{H-NMR}$ analysis, no formation of a cis isomer was admitted and a diastereomer ratio of the trans isomer was 90/10. Yield based on 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 91%. trans isomer (diastereomer mixture)

MS (EI) m/z : 419 (M^+), 155.

IR (KBr, cm^{-1}): 1757 (s), 1504 (m), 1399 (m).

Elemental analysis:			
Calcd:	C, 68.73;	H, 5.05;	N, 3.34
Found:	C, 68.09;	H, 5.05;	N, 3.34.

Calcd:	C, 68.73;	H, 5.05;	N, 3.34
Found:	C, 68.09;	H, 5.05;	N, 3.34.

$^1\text{H-NMR}$ (δ , CDCl_3) (major isomer): 1.82 (d, $J=6.8$ Hz, 3H), 3.22 (d, $J=4.8$ Hz, 1H), 3.72 (s, 3H), 5.09 (d, $J=4.8$ Hz, 1H), 5.84 (s, 2H), 5.99 (q, $J=6.8$ Hz, 1H), 6.23-7.97 (m, 10H), (minor isomer): 1.70 (d, $J=6.8$ Hz, 3H), 2.59 (s, 3H), 4.08 (d, $J=4.8$ Hz, 1H), 5.84 (s, 2H), 5.85 (d, $J=4.8$ Hz, 1H), 6.09 (q, $J=6.8$ Hz, 1H), 6.23-7.97 (m, 10H).

Example 48: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone

[0201] Under argon atmosphere, in 3 ml of methylene chloride were dissolved 134.7 mg (0.5 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 75.1 mg (0.5 mmol) of 3,4-methylenedioxybenzaldehyde, and after the

solution was cooled to -78°C , 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M-boron trifluoride diethyl etherate was added to the solution and the mixture was reacted under stirring for 2.5 hours.

[0202] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 182.2 mg (0.43 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone as pale yellowish crystal. According to the $^1\text{H-NMR}$ analysis, no formation of a cis isomer was admitted and a diastereomer ratio of the trans isomer was 79/21. Yield based on 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 87%.

Example 49: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone

[0203] Under argon atmosphere, in 3 ml of methylene chloride were dissolved 134.7 mg (0.5 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 75.1 mg (0.5 mmol) of 3,4-methylenedioxybenzaldehyde, and after the solution was cooled to -78°C , 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M-t-butyldimethylsilyl trifluoromethanesulfonate was added to the solution and the mixture was reacted under stirring for 30 minutes.

[0204] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, and dried over anhydrous magnesium sulfate to obtain 195.2 mg (0.47 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone as pale yellowish crystal. According to the $^1\text{H-NMR}$ analysis, no formation of a cis isomer was admitted and a diastereomer ratio of the trans isomer was 88/12. Yield based on 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 93%.

Example 50: Synthesis of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone

[0205] Under argon atmosphere, in 3 ml of methylene chloride were dissolved 134.7 mg (0.5 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone and 75.1 mg (0.5 mmol) of 3,4-methylenedioxybenzaldehyde, and after the solution was cooled to -78°C , 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M triisopropylsilyl trifluoromethanesulfonate was added to the solution and the mixture was raised and reacted at -20°C for 1.5 hours.

[0206] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, and dried over anhydrous magnesium sulfate to obtain 203.2 mg (0.48 mmol) of 3-((R)-1-(1-naphthyl)ethyl)-4-methoxycarbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone as pale yellowish crystal. According to the $^1\text{H-NMR}$ analysis, no formation of a cis isomer was admitted and a diastereomer ratio of the trans isomer was 75/25. Yield based on 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone was 97%.

Example 51: Synthesis of 3-((S)-1-phenylethyl)-4-((*l*)-menthyloxy)carbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone

[0207] Under argon atmosphere, in 3 ml of methylene chloride were dissolved 171.7 mg (0.5 mmol) of 3-((S)-1-phenylethyl)-5-((*l*)-menthyloxy)-2(3H)oxazolone and 75.1 mg (0.5 mmol) of 3,4-methylenedioxybenzaldehyde, and after the solution was cooled to -78°C , 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M-trimethylsilyl tri-
 45
 flate was added to the solution and the mixture was reacted under stirring for 30 minutes.

[0208] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 214 mg (0.43 mmol) of 3-((S)-1-phenylethyl)-4-((*l*)-menthyloxy)carbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone as pale yellowish viscous liquid. According to the $^1\text{H-NMR}$ analysis, no formation of a cis isomer was admitted and a diastereomer ratio of a trans isomer was 70/30. Yield based on 3-((S)-1-phenylethyl)-5-((*l*)-menthyloxy)-2(3H)oxazolone was 87%.

trans isomer (diastereomer mixture)

MS (EI) m/z : 493 (M^+), 105.

$^1\text{H-NMR}$ (δ , CDCl_3) (major isomer): 0.65-2.04 (m, 21H), 3.75 (d, $J=4.4\text{Hz}$, 1H), 4.76 (ddd, $J=4.4\text{Hz}$, 1H), 5.14 (d,

J=4.4Hz, 1H), 5.32 (q, J=7.2Hz, 1H), 5.94 (s, 1H), 6.45-7.41 (m, 9H), (minor isomer): 0.65-2.04 (m, 21H), 4.66 (d, J=4.4Hz, 1H), 4.56 (ddd, J=4.4Hz, 1H), 4.92 (q, J=7.2Hz, 1H), 5.15 (d, J=4.4Hz, 1H), 5.99 (s, 1H), 7.09-7.39 (m, 9H).

- 5 Example 52: Synthesis of 3-((S)-1-phenylethyl)-4-((R)-menthyloxy)carbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone

[0209] Under argon atmosphere, in 3 ml of methylene chloride were dissolved 171.7 mg (0.5 mmol) of 3-((S)-1-phenylethyl)-5-((R)-menthyloxy)-2(3H)oxazolone and 75.1 mg (0.5 mmol) of 3,4-methylenedioxybenzaldehyde, and after the solution was cooled to -78°C, 0.5 ml (0.05 mmol) of a methylene chloride solution containing 0.1 M triisopropylsilyl trifluoromethanesulfonate was added to the solution and the temperature of the mixture was raised to room temperature to react them under stirring for one hour.

[0210] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 170.5 mg (0.35 mmol) of 3-((S)-1-phenylethyl)-4-((R)-menthyloxy)carbonyl-5-(3,4-methylenedioxyphenyl)-2-oxazolidinone as pale yellowish viscous liquid. According to the ¹H-NMR analysis, a diastereomer ratio of a trans isomer was 51/49. Yield based on 3-((S)-1-phenylethyl)-5-((R)-menthyloxy)-2(3H)oxazolone was 69%.

Example 53: Synthesis of 3-phenyl-4-methyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone

[0211] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 102.6 mg (0.5 mmol) of 3-phenyl-4-methyl-5-methoxy-2(3H)oxazolone, and after the solution was cooled to -78°C, 53.1 mg (0.5 mmol) of benzaldehyde and 0.5 in 1 (0.05 mmol) of a methylene chloride solution containing 0.1 M-trimethylsilyl triflate were added to the solution and the mixture was reacted under stirring for 1.0 hour. After the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. The obtained concentrated residue was applied to silica gel column chromatography (eluent, hexane:ethyl acetate = 10:1 (volume ratio)) to obtain 146.8 mg (0.47 mmol) of diastereomer mixture crude product of 3-phenyl-4-methyl-4-methoxycarbonyl-5-phenyl-2-oxazolidinone as pale yellowish liquid. A diastereomer ratio thereof was 72/28 according to ¹H-NMR. Yield based on 3-phenyl-4-methyl-5-methoxy-2(3H)oxazolone was 94.3%.

(Mixture)

MS (EI) m/z: 311 (M⁺), 252, 208, 118.

¹H-NMR (δ, CDCl₃) (major isomer): 1.04 (s, 3H), 3.89 (s, 3H), 5.74 (s, 1H), 7.26-7.44 (m, 10H).

(minor isomer): 1.72 (s, 3H), 3.35 (s, 3H), 5.42 (s, 1H), 7.26-7.44 (m, 10H).

- 40 Example 54: Synthesis of 3-benzyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-isobutyl-2-oxazolidinone

[0212] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 202.8 mg (0.5 mmol) of 3-benzyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)-oxazolone, and after the solution was cooled to -78°C, 36.1 mg (0.5 mmol) of isobutyl aldehyde and 63 μl (0.1 mmol) of boron trifluoride diethyl etherate were added to the solution and the mixture was reacted for 1.0 hour. After the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 232.1 mg (0.49 mmol) of a 4,5-trans isomer alone of 3-benzyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-isobutyl-2-oxazolidinone as milky white viscous liquid. The obtained trans isomer was confirmed to be a single diastereomer according to ¹H-NMR. Yield based on 3-benzyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone was 97.2%.

MS (CI, i-C₄H₁₀) m/z: 478 (MH⁺).

¹H-NMR (δ, CDCl₃): 0.62-2.01 (m, 25H), 2.95 (d, J=4.88Hz, 1H), 3.82 (d, J=4.88Hz, 1H), 4.18 (d, Jgem=15.13Hz, 1H), 4.82 (ddd, J=3.91Hz, 1H), 4.99 (d, Jgem=15.13Hz, 1H), 6.82-7.45 (m, 10H)

Example 55: Synthesis of 3-benzyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-(2-phenylethyl)-2-oxazolidinone

[0213] Under argon atmosphere, in 3 ml of methylene chloride was dissolved 202.8 mg (0.5 mmol) of 3-benzyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)-oxazolone, and after the solution was cooled to -78°C, 67.1 mg (0.5 mmol) of hydrocinnamaldehyde and 63 µl (0.1 mmol) of boron trifluoride diethyl etherate were added to the solution and the mixture was reacted for 1.0 hour. After the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 15 ml of methylene chloride. The organic layer was washed twice with 15 ml of water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 264.9mg (0.49 mmol) of 3-benzyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-(2-phenylethyl)-2-oxazolidinone as pale yellowish viscous liquid. A formation ratio of cis/trans isomers was cis isomer/trans isomer = 29/71 according to ¹H-NMR, and they were each single diastereomer. Yield based on 3-benzyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)2(3H)oxazolone was 98.2%.

MS (CI, i-C₄H₁₀) m/z: 540 (MH⁺).

¹H-NMR (δ, CDCl₃): (trans isomer) 0.60-2.04 (m, 20H), 2.55-2.58 (m, 2H), 2.81 (d, J=5.86Hz, 1H), 3.93 (d, J=5.86Hz, 1H), 4.23 (d, Jgem=15.14Hz, 1H), 4.74 (ddd, J=3.91Hz, 1H), 4.98 (d, Jgem=15.14Hz, 1H), 6.78-7.45 (m, 15H); (cis isomer) 0.60-2.04 (m, 20H), 2.58 (d, J=8.79Hz, 1H), 3.22 (d, J=8.79Hz, 1H), 3.75-4.13 (m, 2H), 4.08 (d, Jgem=15.14Hz, 1H), 4.14 (ddd, J=3.91Hz, 1H), 5.15 (d, Jgem=15.14Hz, 1H), 6.78-7.45 (m, 15H).

Example 56: Synthesis of 3-diphenylmethyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-phenyl-2-oxazolidinone

[0214] In 4 ml of methylene chloride were dissolved of 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone (0.24 g; 0.5 mmol) and benzaldehyde (0.053 g; 0.5 mmol), and after the solution was cooled to -78°C, 15 mg of triisopropylsilyl triflate was added to the solution and the mixture was reacted under stirring at -78°C for 3 hours.

[0215] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 0.30 g of 3-diphenylmethyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-phenyl-2-oxazolidinone as yellowish crystal. According to the ¹H-NMR analysis, a formation ratio of cis/trans isomers of the formed product was 74/26. Also, a cis isomer and a trans isomer were each obtained as a single diastereomer. Yield based on 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)-oxazolone was 99%.

[0216] The spectrum data were the same as those obtained in Example 28.

Example 57: Synthesis of 3-diphenylmethyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-(2-phenylethyl)-2-oxazolidinone

[0217] In 4 ml of methylene chloride were dissolved of 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone (0.24 g; 0.5 mmol) and hydrocinnamaldehyde (0.067 g; 0.5 mmol), and after the solution was cooled to -78°C, 10 mg of BF₃ · Et₂O was added to the solution and the mixture was reacted under stirring, at -78°C for 2 hours.

[0218] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 0.29 g of 3-diphenylmethyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)carbonyl-5-(2-phenylethyl)-2-oxazolidinone as white crystal. According to the ¹H-NMR analysis, a formation ratio of cis/trans isomers of the formed product was 13/87. Also, a cis isomer and a trans isomer were each obtained as a single diastereomer. Yield based on 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone was 94%.

MS (EI) m/z: 615 (MH⁺), 167.

cis isomer

¹H-NMR (δ, CDCl₃): 2.74-2.90 (m, 2H), 3.51 (d, J=8.3Hz, 1H), 4.27 (m, 1H), 4.52 (td, J=3.9Hz, J=10.7Hz, 1H), 5.90 (s, 1H).

trans isomer

¹H-NMR (δ, CDCl₃): 2.50-2.71 (m, 2H), 3.06 (d, J=5.9Hz, 1H), 3.91 (m, 1H), 4.61 (td, J=3.9Hz, J=10.7Hz, 1H), 5.88 (s, 1H).

Example 58: Synthesis of 3-diphenylmethyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxycarbonyl)-5-isopropyl-2-oxazolidinone

[0219] In 4 ml of methylene chloride were dissolved of 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone (0.24 g; 0.5 mmol) and isobutyl aldehyde (0.036 g; 0.5 mmol), and after the solution was cooled to -78°C, 10 mg of BF₃ · Et₂O was added to the solution and the mixture was reacted under stirring at -78°C for 5 hours.

[0220] Moreover, after the temperature of the mixture was raised up to the room temperature, 15 ml of a saturated aqueous sodium hydrogen carbonate solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 0.27 g of 3-diphenylmethyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxycarbonyl)-5-isopropyl-2-oxazolidinone as white crystal. According to the ¹H-NMR analysis, a formation ratio of cis/trans isomers of the formed product was 81/19. Also, a cis isomer and a trans isomer were each obtained as a single diastereomer. Yield based on 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone was 97%.

MS (EI) m/z: 553 (M⁺), 167.

cis isomer

¹H-NMR (δ, CDCl₃): 3.65 (d, J=7.3Hz, 1H), 4.04 (t, J=7.3Hz, 1H), 4.57 (td, J=10.7Hz, J=3.9Hz, 1H), 5.94 (s, 1H).

trans isomer

¹H-NMR (δ, CDCl₃): 3.23 (d, J=4.9Hz, 1H), 3.86 (d, J=4.9Hz, 1H), 4.76 (td, J=4.3Hz, J=10.7Hz, 1H), 5.78 (s, 1H).

Example 59: Synthesis of 3-diphenylmethyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxycarbonyl)-5-phenyl-2-oxazolidinone

[0221] In 4 ml of methylene chloride were dissolved of 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone (0.24 g; 0.5 mmol) and benzaldehyde (0.053 g; 0.5 mmol), and after the solution was cooled to -78°C, 0.06 ml of a hexane solution containing 0.9 M diethyl aluminum chloride was added to the solution and the mixture was reacted under stirring at -78°C for 2 hours.

[0222] Moreover, after the temperature of the mixture was raised up to the room temperature and the mixture was reacted for 3 hours, 15 ml of an aqueous 0.5N-hydrochloric acid solution was added to the resulting reaction mixture and the mixture was extracted with 20 ml of methylene chloride. The organic layer was washed with a saturated aqueous sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 0.27 g of 3-diphenylmethyl-4-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxycarbonyl)-5-phenyl-2-oxazolidinone as white crystal. According to the ¹H-NMR analysis, a formation ratio of cis/trans isomers of the formed product was 74/26. Also, a cis isomer and a trans isomer were each obtained as a single diastereomer. Yield based on 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone was 90%.

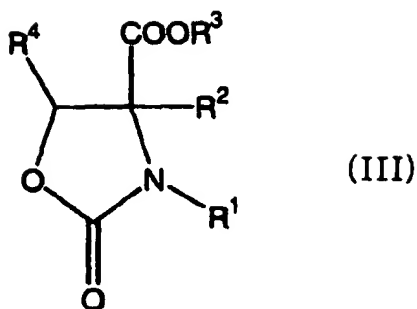
[0223] The spectrum data were the same as those obtained in Example 28.

Utilizability in industry

[0224] According to the present invention, a 4-alkoxycarbonyl-2-oxazolidinone compound can be obtained by reacting 5-alkoxy-2(3H)oxazolone compound and an aldehyde compound in the presence of a Lewis acid catalyst. The obtained 4-alkoxycarbonyl-2-oxazolidinone compound is useful as a starting material of a β-hydroxy-α-amino acid compound which has been used as a drug matter, an intermediate or a starting material.

Claims

1. A process for producing a 4-alkoxycarbonyl-2-oxazolidinone compound represented by the formula (III):



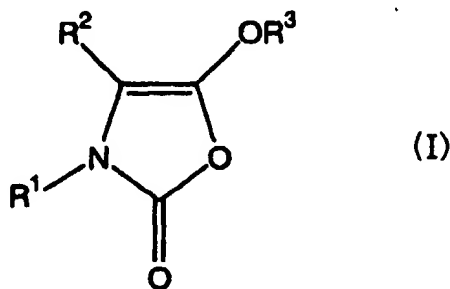
15

wherein R¹ represents a hydrogen atom, a C₁ to C₁₀ alkyl group which may be substituted, a C₃ to C₁₀ cycloalkyl group which may be substituted, a C₂ to C₁₀ alkenyl group which may be substituted or a phenyl group which may be substituted, R² represents a hydrogen atom, a C₁ to C₁₀ alkyl group which may be substituted, a phenyl group which may be substituted or a C₂ to C₁₀ alkenyl group which is not substituted, R³ represents a C₁ to C₁₀ alkyl group which may be substituted, a C₃ to C₁₀ cycloalkyl group which may be substituted, a C₂ to C₁₀ alkenyl group which may be substituted (provided that a 2-alkenyl group is excluded), or a phenyl group which may be substituted, and R⁴ represents a hydrogen atom, a C₁ to C₂₀ alkyl group which may be substituted, a C₂ to C₂₀ alkenyl group which may be substituted, a C₃ to C₁₀ cycloalkyl group which may be substituted, a C₂ to C₂₀ alkynyl group which may be substituted, a C₆ to C₂₀ aryl group which may be substituted, a 5- or 6-membered heteroaromatic ring group having 1 or 2 hetero atoms selected from N, O and S which may be substituted, a C₁ to C₆ alkoxycarbonyl group which may be substituted, an acetyl group or a benzoyl group,

20

25

30 which comprises reacting a 5-alkoxy-2(3H)oxazolone compound represented by the formula (I):



45 wherein R¹, R² and R³ have the same meanings as defined above,

and an aldehyde compound represented by the formula (II):



wherein R⁴ has the same meaning as defined above,

in the presence of a Lewis acid catalyst.

55

2. The method according to Claim 1, wherein the Lewis acid is a halide or a trifluoromethan sulfonate of an element from Group 2 (Group IIa) to Group 14 (Group IVa) of the Periodic Table (except for carbon), or a Lanthanoid group metal.

3. The method according to Claim 1, wherein the Lewis acid is a compound represented by the formula (IV):



wherein R^5 represents a C_1 to C_{10} alkyl group or a C_6 to C_{20} aryl group; X represents a halogen atom; M represents Al, B, Sn or Ti; m and n each represents a number of 0, 1, 2, 3 or 4; provided that $m + n$ is 2, 3 or 4.

4. The method according to Claim 1, wherein the Lewis acid is a compound represented by the formula (V):



wherein R^6 represents a C_1 to C_{10} alkyl group or a C_6 to C_{20} aryl group; X represents a halogen atom; M represents Al, B, Sn or Ti; m' and n' each represents a number of 0, 1, 2, 3 or 4; provided that $m' + n'$ is 3 or 4.

5. The method according to Claim 1, wherein the Lewis acid is a compound represented by the formula (VI):



wherein R^7 , R^8 and R^9 each independently represents a C_1 to C_{10} alkyl group or a C_6 to C_{20} aryl group; X represents a halogen atom or $-OSO_2CF_3$.

6. The method according to Claim 1, wherein the Lewis acid is aluminum (III) chloride, aluminum (III) bromide, aluminum (III) iodide, diethyl aluminum chloride, ethyl aluminum dichloride, trimethyl aluminum, triethyl aluminum, triisopropyl aluminum, tributyl aluminum, aluminum (III) isopropoxide, boron trichloride, boron trifluoride, boron trifluoride diethyl etherate, boron tribromide, triphenoxyborane, phenyl dichloroborane, tin (IV) chloride, tin (IV) bromide, tin (II) chloride, tin (II) triflate, titanium (IV) chloride, titanium (IV) fluoride, titanium (IV) bromide, titanium (IV) iodide, dichloroisopropoxy titanium, titanium (IV) isopropoxide, zinc (II) chloride, zinc (II) bromide, iron (II) chloride, iron (III) chloride, magnesium chloride, ytterbium (III) triflate, samarium iodide, samarium (III) triflate, trimethylsilyl triflate, trimethylsilyl iodide, tert-butyldimethylsilyl triflate or triisopropylsilyl triflate.

7. The method according to Claim 1, wherein the compound represented by the formula (I) is 3-benzyl-5-(ℓ)-menthyloxy-2(3H)oxazolone, 3-benzyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone, 3-(4-methylbenzyl)-4-methyl-5-methoxy-2(3H)oxazolone, 3-(1-phenylethyl)-5-methoxy-2(3H)oxazolone, 3-((S)-1-phenylethyl)-5-isopropoxy-2(3H)oxazolone, 3-((R)-1-phenylethyl)-5-methoxy-2(3H)oxazolone, 3-diphenylmethyl-5-methoxy-2(3H)oxazolone, 3-diphenylmethyl-5-(ℓ)-menthyloxy-2(3H)oxazolone, 3-diphenylmethyl-5-((1S,2S,5R)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyloxy)-2(3H)oxazolone, 3-((R)-1-(1-naphthyl)ethyl)-5-methoxy-2(3H)oxazolone, 3-furfuryl-4-ethyl-5-methoxy-2(3H)oxazolone, 3-furfuryl-4-ethyl-5-(4-pentenyl)oxy-2(3H)oxazolone, 3-isopropyl-5-methoxy-2(3H)oxazolone, 3-benzyl-5-methoxy-2(3H)oxazolone, 3-isopropyl-4-methyl-5-methoxy-2(3H)oxazolone, 3-isopropyl-4-methyl-5-ethoxy-2(3H)oxazolone, 3-isopropyl-4-methyl-5-cyclohexyloxy-2(3H)oxazolone, 3-((R)-1-(1-naphthyl)ethyl)-5-isopropoxy-2(3H)oxazolone, 3-((R)-1-phenylethyl)-4-methyl-5-methoxy-2(3H)oxazolone, 3-((S)-1-phenylethyl)-5-phenoxy-2(3H)oxazolone, 3-benzyl-4-methyl-5-methoxy-2(3H)oxazolone, 3-(1-naphthyl)methyl-5-methoxy-2(3H)oxazolone, 3-((R)-1-phenylethyl)-5-(ℓ)-menthyloxy-2(3H)oxazolone, 3-((S)-1-phenylethyl)-5-(ℓ)-menthyloxy-2(3H)oxazolone or 3-phenyl-4-methyl-5-methoxy-2(3H)oxazolone.

8. The method according to Claim 1, wherein the compound represented by the formula (II) is benzaldehyde, p-methoxybenzaldehyde, o-methoxybenzaldehyde, 3,4-methylenedioxybenzaldehyde, 3,4-dimethoxybenzaldehyde, p-nitrobenzaldehyde, o-nitrobenzaldehyde, naphthylaldehyde, acetaldehyde, propionaldehyde, n-butylaldehyde, isobutyl aldehyde, cinnamaldehyde, hydrocinnamaldehyde, crotonaldehyde, phenylacetaldehyde, α -benzyloxypropionaldehyde, methylglycidate, acrolein, tetradecenal, or benzyloxyacetaldehyde.

INTERNATIONAL SEARCH REPORT

International application No.
 PCT/JP98/02129

A. CLASSIFICATION OF SUBJECT MATTER Int.Cl ⁶ C07D263/24, 263/44		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) Int.Cl ⁶ C07D263/24, 263/44		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) REGISTRY (STN), CAPLUS (STN)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
E, X	WO, 98/22449, A1 (Ube Industries, Ltd.), May 28, 1998 (28. 05. 98), (Refer to "Background Arts") (Family: none)	1-8
A	US, 5384412, A (The Scripps Research Institute), January 24, 1995 (24. 01. 95), (Refer to Figure 7, step a, Example 31) (Family: none)	1-8
A	JP, 60-34955, A (Showa Denko K.K.), February 22, 1985 (22. 02. 85), (Refer to Claims) (Family: none)	1-8
A	JP, 1-228946, A (Suntory Ltd.), September 12, 1989 (12. 09. 89), (Refer to Claims ; Example 4) (Family: none)	1-8
<input checked="" type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family		
Date of the actual completion of the international search August 5, 1998 (05. 08. 98)		Date of mailing of the international search report August 18, 1998 (18. 08. 98)
Name and mailing address of the ISA/ Japanese Patent Office		Authorized officer
Facsimile No.		Telephone No.

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INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP98/02129

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US, 4186129, A (Schering Aktiengesellschaft), January 29, 1980 (29. 01. 80), (Refer to EXAMPLE 20) & DE, 2655369, A1 & FR, 2372814, A1 & JP, 53-82777, A	1-8

Form PCT/ISA/Z10 (continuation of second sheet) (July 1992)